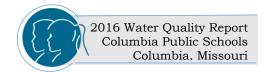


Abstract

Established in 1873, Columbia Public Schools has a proud history of serving the city of Columbia, Missouri. With a history spanning more than 143 years, the district has gained a reputation as an educational pioneer for developing best practices and innovative programs to support its mission to provide an excellent education for all students. More than 18,000 students now attend any one of the 33 schools in the district.

Due to the recent heightened awareness of lead in consumable water; Columbia Public Schools has elected to be proactive and conduct a water quality evaluation for all schools and support facilities in the district. The evaluation is believed to be the first of its kind by a school district in the State of Missouri. But more importantly, it provides a direction for students, parents, teachers and general public on issues where lead in water may be a concern at their school.

Ultimately keeping with the Columbia Public Schools Facility & Construction Services safety motto, Remember, Kids Learn Best in Healthy Environments...



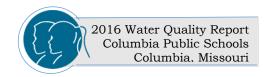
1.0 Executive Summary

On June 3, 2016 (initial water quality evaluation) and July 7, 2016 (follow-up water quality evaluation) Columbia Public Schools (CPS) Facilities and Construction Services Department (FCS) provided assistance for a water quality evaluation of West Middle School, 401 Clinkscales Road, Columbia, Missouri 65203.

PDC Laboratories Inc., (PDC) was contracted by CPS to provide the following services for the water quality evaluation. PDC Laboratories, Inc. role included: (1) transporting the water samples from the FCS location to their laboratory location, (2) insuring the field data form(s) were signed, accurate, and true, (3) conducting the analysis of each water sample that was collected at the school, (4) providing an analytical laboratory report for each sample documented on the school's field data form; and (5) developing a summary report for each school.

The initial summary report is divided into three (3) parts or sections;

- (1) PART A of the report shows the total number of samples that were collected at the school. Also, a percentage is given for the total number of samples (lead and/or copper) that were below the lead and/or copper action limit(s). The passing requirement noted in the United States Environmental Protection Agency (USEPA) Lead and Copper Rule (LCR) for public water supplies is 90%.
- (2) PART B of the report shows the number of samples that have exceeded the lead and/or copper action limit. The USEPA lead and copper action level (lead action limit = 15 UG/L; copper action limit = 1,300 UG/L).
- (3) Conclusions/Comments of report discusses the findings of the school and how it is compared to the (recommended) guidelines of the USEPA-LCR and the Columbia Water & Light requirements. Additionally, it is in this section where PDC provides suggestions (options) for any laboratory sample that has exceeded the lead and/or copper level.



The follow up summary report is divided into three (3) parts or sections;

- (1) Follow up Results of the report shows the laboratory result(s) of the sample.
- (2) Follow up samples passed action limits of the report simply defines the sample as "yes" or "no". The USEPA lead and copper action level (lead action limit = 15 UG/L; copper action limit = 1,300 UG/L).
- (3) Conclusions/Comments of report discusses the recommendations (typically noting that the location should be flushed for a minimum of 30 seconds prior to use) and how it is compared to the (recommended) guidelines of the USEPA-LCR and the Columbia Water & Light requirements.

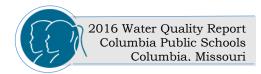
Initial Water Quality Evaluation

On June 3, 2016, a total of 53 (potential) water consumable samples were collected at various water locations within the school.

The following data table contains the water sample number, description/location of the collected sample, and the response action performed by CPS Facilities & Construction Services (CPS-FCS), if needed.

West Middle School

Sample		CPS-FCS
No.	Description/Location	Actions
1	Kitchen Faucet (cold) in center	0
2	Kitchen Faucet (cold) in steam kettle	0
3	Kitchen Faucet (cold) in 5-bay sink	0
4	Family and Consumer Science Faucet (cold) in room 229	0
5	Family and Consumer Science Faucet (cold) in room 229	0
6	Family and Consumer Science Faucet (cold) in room 229	0
7	Electrical Water Cooler in hall outside room 227	0
8	Teacher's Lounge Faucet (cold) in room 221	0
9	Electrical Water Cooler in hall outside room 222	0
10	Nurse's Sink Faucet (cold) in room 222	0
11	Electrical Water Cooler in hall outside room 212	0
12	Electrical Water Cooler in hall outside room 211	0
13	Electrical Water Cooler in hall outside room 107	0
14	Electrical Water Cooler in hall outside room 112	0
15	Electrical Water Cooler in hall outside room 123	0
16	Electrical Water Cooler in hall outside room 101	0
17	Electrical Water Cooler in hall outside room 136	0
18	Electrical Water Cooler in room 136L	0



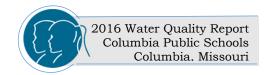
Sample No.	Description/Location	CPS-FCS Actions
19 20 21	Electrical Water Cooler in room 137L Electrical Water Cooler in room 135A Electrical Water Cooler in hall outside room 125	0 0 0
22	Electrical Water Cooler in hall outside room 130	0
23	Electrical Water Cooler in room 132	0
24	Electrical Water Cooler in room 133	0
25	Classroom Faucet (cold) in room 131	0
26	Classroom Faucet (cold) in room 131	1
27	Classroom Faucet (cold) in room 131	0
28	Classroom Faucet (cold) in room 124	0
29	Classroom Faucet (cold) in room 124	0
30	Classroom Faucet (cold) in room 124	0
31	Classroom Faucet (cold) in room 1250	0
32	Classroom Faucet (cold) in room 125	0
33 34	Classroom Faucet (cold) in room 125	0 0
34 35	Classroom Faucet (cold) in room 126	0
36	Classroom Faucet (cold) in room 126 Classroom Faucet (cold) in room 126	0
37	Classroom Faucet (cold) in room 1260	0
38	Classroom Faucet (cold) in room 127	0
39	Classroom Faucet (cold) in room 127	0
40	Classroom Faucet (cold) in room 128	Ö
41	Classroom Faucet (cold) in room 128	Ö
42	Classroom Faucet (cold) in room 128	0
43	Classroom Faucet (cold) in room 1280	0
44	Classroom Faucet (cold) in room 129	0
45	Classroom Faucet (cold) in room 129	0
46	Classroom Faucet (cold) in room 129	0
47	Classroom Faucet (cold) in room 130	0
48	Classroom Faucet (cold) in room 130	0
49	Classroom Faucet (cold) in room 130	0
50	Classroom Faucet (cold) in room 1300	0
51	Classroom Faucet (cold) in room 111	0
52	Classroom Faucet (cold) in room 1235	1
53	Classroom Faucet (cold) in room 123	1

CPS-FCS Actions

0 = No actions are required. The laboratory results revealed that the samples have no concerns for lead and/or copper.

1 = CPS-FCS & PDC Laboratories recommend(s) the following actions:

(1) An inspection of the water tap and associated water line by a CPS-FCS specialized maintenance plumber to help determine the cause, (2) CPS-FCS shall conduct a follow-up sample collection-a flush sample*, and (3) CPS-FCS shall post a sign saying "Notice" "work station sink, water not intended for drinking." (if needed).



*Follow-up Sample(s)

This sample is representative of the water that is in the plumbing upstream from the faucet or drinking water fountain. This sample shall be collected before the facility opens and before any water was used at the water tap. The water from the faucet or drinking water fountain should run for 30 seconds before collecting the sample. This is called a flush sample.

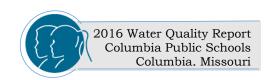
Part A of the PDC summary report supports that 100% of all the water samples collected/analyzed were at a none detect and/or below the EPA copper action level (copper action limit = 1,300 UG/L). Additionally, the PDC summary report supports that 94.3% of all the water samples collected/analyzed were at a none detect and/or below the EPA lead action level (lead action limit = 15 UG/L).

Part B of the PDC summary report shows that none (0) number of the water samples collected/analyzed exceeded the EPA copper action level. Additionally, the PDC summary report shows that three (3) of the water samples collected/analyzed exceeded the EPA lead action level.

The PDC conclusions/comments section of the summary report suggest that if any of the laboratory analysis reveals that the sample has exceeded the action limit(s) then the options in the LCR might be considered. They include but are not limited to: flushing prior to use, repairing or replacing the fixture, repairing or replacing the piping, or by simply not using the water for consumption. All of these suggestions are recommended by the LCR.

Based on the PDC conclusions/comments and the CPS-FCS team approach to insure that the water quality evaluation had/has achieved its objective (*Kids Learn Best in Healthy Environments*) to provide safe drinking water in the Columbia Public Schools, CPS-FCS had decided to take these options:

- 1. Taking a second sample (as per EPA LCR and the Drinking Water Best Management Practices suggestive guidelines) of the water source,
- 2. Having a CPS-FCS specialized maintenance plumber inspect the water tap and pipe to make sure there are no leaks or outside indications of corrosion, and



3. Post a sign saying "Notice" "work station sink, water not intended for drinking. " (to be determined after the additional testing result and summary recommendation, if needed).

Follow-up Water Quality Evaluation

On July 7, 2016, CPS-FCS performed a flush sample (follow up water sample) of each of the three water samples that had exceeded the lead action limit on the initial water quality evaluation. Additionally, a CPS-FCS specialized maintenance plumber inspected the water tap and associated pipe to make sure there were no leaks or outside indications of corrosion.

The following data table contains the laboratory reporting and summary for each of the three samples:

SAMPLE 1

Classroom faucet in room 131

Sample Code: WMS-FS-CF-2

Follow up Results:

Lead 1.2 ug/l

Copper 85 ug/l

Follow up samples passed action limits:

Lead (action limit 15 ug/l) YES

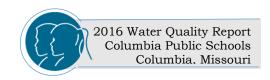
Copper (action limit 1300 ug/l) YES

Part 1 of the PDC follow up sample-flush sample summary report supports that the water sample collected/analyzed was at a none detect and/or below the EPA copper and lead action level (copper action limit = 1,300 UG/L, lead action limit = 15 UG/L).

Part 2 of the PDC summary report shows that the sample(s) meets (passed) the EPA lead and copper action levels.

PDC CONCLUSIONS/COMMENTS

The water tap (plus all other water taps within the rooms) can be used and should be flushed for a minimum of 30 seconds prior to use.



SAMPLE 2

Classroom faucet in room 123

Sample Code: WMS-FS-1235-CF-28

Follow up Results:

Lead 18 ug/l

Copper 150 ug/l

Follow up samples passed action limits:

Lead (action limit 15 ug/l) NC

Copper (action limit 1300 ug/l) YES

Part 1 of the PDC follow up sample-flush sample summary report supports that the water sample collected/analyzed was at a none detect and/or below the EPA copper action limit (copper action limit = 1,300 UG/L). However, the PDC follow up sample-flush sample summary report supports that the water sample collected/analyzed was <u>above</u> the lead action limit (lead action limit = 15 UG/L).

Part 2 of the PDC summary report shows that the sample meets (passed) the EPA copper action level. However, the PDC summary report shows that the sample does not meet (elevated) the EPA lead action level.

PDC CONCLUSIONS/COMMENTS

The water tap (plus all other water taps within the rooms) should not be used for consumption purposes, and should be labelled as such. As an immediate precaution, CPS-FCS has discontinued the water service to the classroom.

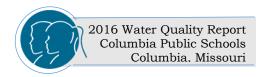
SAMPLE 3

(1) Classroom faucet in room 123

Sample Code: WMS-FS-123-CF-29

Follow up Results:

Lead 22 ug/l



Copper 82 ug/l

Follow up samples passed action limits:

Lead (action limit 15 ug/l) NO

Copper (action limit 1300 ug/l) YES

Part 1 of the PDC follow up sample-flush sample summary report supports that the water sample collected/analyzed was at a none detect and/or below the EPA copper action limit (copper action limit = 1,300 UG/L). However, the PDC follow up sample-flush sample summary report supports that the water sample collected/analyzed was <u>above</u> the lead action limit (lead action limit = 15 UG/L).

Part 2 of the PDC summary report shows that the sample meets (passed) the EPA copper action level. However, the PDC summary report shows that the sample does not meet (elevated) the EPA lead action level.

PDC CONCLUSIONS/COMMENTS

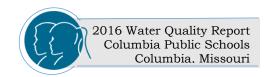
The water tap (plus all other water taps within the rooms) should not be used for consumption purposes, and should be labelled as such. As an immediate precaution, CPS-FCS has discontinued the water service to the classroom.

The PDCs initial water quality summary report and the follow up water quality summary can be found in Appendix A of this report

WATER QUALITY EVALUATION CONCLUSIONS

The conclusions of the water quality evaluation(s) are based herein and were formed from:

- The PDC Laboratories Inc., Summary (Appendix A);
- The PDC Laboratories Inc., Follow-up Summary Report (Appendix A);
- The PDC Laboratories Inc., Laboratory Analytical Results (Appendix B); and
- The most current guidelines of the USEPA-LCR, Drinking Water Best Management Practices and the Columbia Water & Light requirements (Appendix D).



Finally, individual drinking water taps should be flushed to rid stagnant water from the pipes. At the start of each day, before using any water for drinking or cooking, flush the cold water faucet by allowing the water to run for a period of time. Flushing times vary depending on the buildings pipes and outlets. Industry experts and CPS-FCS suggests a time of 30 seconds to 2 minutes be used to properly flush out the reservoir.

As of July 29, 2016, West Middle School meets the (recommended) requirements of the United States Environmental Protection Agency (USEPA) and the published Lead and Copper Rule (LCR) of 1991 (reference: 40 CFR Part 141 Subpart I). Public Run Water Suppliers must comply with the USEPA law and the LCR. Columbia Water & Light is the publicly run water utility that serves this school. Although the Lead and Copper Rule does not apply directly to Columbia Public Schools and each of its schools or support buildings that were part of the 2016 Water Quality evaluation; it is a common practice for experts in the industry to use the standard as guideline when conducting water assessment studies.

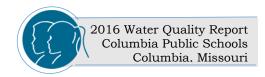
Appendix A

Appendix A of this report includes the PDC Laboratories Inc., summary and follow up summary of the school. The summaries provide the analytical results, interpretation of the results and recommended actions based on the LCR.

PDC Laboratories, Inc. role which involved: (1) transporting the water samples from the FCS location to their laboratory location, (2) insuring the field data form(s) were signed, accurate and true, (3) conducting the analysis of each water sample that was collected at the school, (4) providing an analytical laboratory report for each sample documented on the school's field data form, and (5) developing a summary report for each school.

The Summary reports include:

- The date of the analysis and the report,
- The school and/or support facility name,
- Number of water samples received from the evaluation,
- Number of samples that are known to be below the copper & lead action levels,
- The action levels (copper & lead) that can be utilized for a comparison,
- The laboratory findings that included analytical results,
- Comparing the analytical results to the LCR (suggested) action levels.
- The number of samples that exceed the LCR (suggested) action levels.
- The percentage of samples (comparing the results to LCR (suggested),
- Conclusions and Recommendations



Noting: if the facility meets the (recommended) requirements of the LCR, Providing response measures for any outlying sample(s) based on the recommendations by the LCR.

Appendix B

Appendix B of this report contains the analytical report(s) and the chain of custody forms.

Appendix C

PDC Laboratories, Inc. is a State of Missouri-Department of Natural Resources approved Drinking Water Laboratory (1050). The PDC laboratory certifications are located in Appendix C.

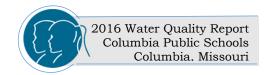
Appendix D

The references are located in Appendix D. References for this evaluation include information from Columbia Water & Light, Columbia, Missouri, the United States Environmental Protection Agency (USEPA) – Lead and Copper Rule, Drinking Water Best Management Practices and the *Clarification of Recommended Tap Sampling Procedures for Purposes of the Lead and Copper Rule (February 29, 2016).*

2.0 Limitations

The findings obtained and our recommendations were prepared in accordance with customary principles and practices in the field of environmental science and engineering. This statement is in lieu of other statements either expressed or implied. This report does not warrant against future operations or conditions, nor does it warrant against operations or conditions present of a type or at a location not investigated. Additionally, the passage of time may result in a change in the environmental characteristics at this school. This report does not warrant against future operations or conditions that could affect the recommendations made.

Water sampling evaluations are limited in the sense that conclusions and recommendations are developed from the United States Environmental Protection Agency (USEPA) Lead and Copper Rule (LCR) of 1991 (reference: 40 CFR Part 141 Subpart I). The LCR requires that all public run water suppliers comply with



the USEPA law. Columbia Water & Light is the publicly run water utility that serves this school.

This report is intended for the sole use of the Columbia Board of Education. The scope of services performed in execution of this evaluation may not be appropriate to satisfy the needs of other users, and use or re-use of this document or the findings, conclusions, or recommendations, is at risk of said user.

Respectfully submitted this 29th Day of July 2016.

CPS-Facilities & Construction Services

Facilities & Construction Services

Appendix A



Facilities & Construction Services

PDC Laboratories Inc.

Summary Report Initial Water Quality Evaluation







COLUMBIA PUBLIC SCHOOLS COLUMBIA, MO LEAD AND COPPER MONITORING SUMMARY REPORT

DATE: June 24, 2016

FACILITY NAME: WEST MIDDLE SCHOOL

PART A	
NUMBER OF SAMPLES COLLECTED AT THIS FACILITY:	53
PERCENTAGE OF SAMPLES BELOW COPPER ACTION LIMIT COPPER ACTION LIMIT = 1,300 UG/L	100.0
PERCENTAGE OF SAMPLES BELOW LEAD ACTION LIMIT LEAD ACTION LIMIT = 15 UG/L	94.3
(The requirement in the USEPA Lead and Copper Rule for public water supplies is 90% passing)	
PART B	0
NUMBER OF SAMPLES EXCEEDING COPPER ACTION LIMIT NUMBER OF SAMPLES EXCEEDING LEAD ACTION LIMIT	0

CONCLUSIONS/COMMENTS

This facility meets the (recommended) requirements of the United States Environmental Protection Agency (USEPA) and the published Lead and Copper Rule(LCR) of 1991 (reference: 40 CFR Part 141 Subpart I). Public Run Water Suppliers must comply with the USEPA law and the LCR. Columbia Water & Light is the puclicly run water utility that serves this school.

Although the Lead and Copper Rule may not directly apply to Columbia Public Schools and each of its schools or support buildings that are part of the 2016 Water Quality Assessment Reporting; it is a common practice for experts in the field and members of the known industry to use the standard as a guideline for water assessment studies.

If any of the laboratory analysis reveals that the sample has exceeded the action limit, then you may want to consider the options as outlined in the LCR. They include but at not limited to: flushing prior to use (make it a practice to run the water at each tap before use), repairing and/or replacing fixtures and/or piping and by not using the water for consumption. All are recommended by the Lead and Copper Rule.

PREPARED BY

KURT C STEPPING SENIOR PROJECT MANAGER PDC LABORATORIES, INC.

Facilities & Construction Services

PDC Laboratories Inc.

Follow-up Sampling Summary Report







COLUMBIA PUBLIC SCHOOLS

SUMMARY REPORT-FOLLOW UP SAMPLE(FLUSH SAMPLE)

DATE: July 18, 2016

FACILITY NAME: WEST MIDDLE SCHOOL

Sample Location:	WMS-FS-131-CF-2*	WMS-FS-1235-CF-28**	WMS-FS-123-CF-29**
Followup Results:			
Lead ug/l	1.2	18	22
Copper ug/l	85	150	82
Followup samples passed action lin	nits:		
Lead(action limit 15 ug/l)	YES	NO	NO
Copper(action limit 1300 ug/l)	YES	YES	YES

CONCLUSIONS/COMMENTS

- *This location should be flushed for a minimum of 30 seconds prior to use.
- ** These locations should not be used for consumption purposes, and should be labelled as such.

This facility meets the (recommended) requirements of the United States Environmental Protection Agency (USEPA) and the published Lead and Copper Rule(LCR) of 1991 (reference: 40 CFR Part 141 Subpart I). Public Run Water Suppliers must comply with the USEPA law and the LCR. Columbia Water & Light is the puclicly run water utility that serves this school.

Although the Lead and Copper Rule may not directly apply to Columbia Public Schools and each of its schools or support buildings that are part of the 2016 Water Quality Assessment Reporting; it is a common practice for experts in the field and members of the known industry to use the standard as a guideline for water assessment studies.

PREPARED BY

KURT C STEPPING SENIOR PROJECT MANAGER PDC LABORATORIES, INC.

Facilities & Construction Services

Appendix B



Facilities & Construction Services

PDC Laboratories Inc. Laboratory Results Initial Water Quality Evaluation

Field Data & Chain-of-Custody Forms





PDC Laboratories, Inc.

PROFESSIONAL • DEPENDABLE • COMMITTED

June 24, 2016

Randy Jackson Columbia Public Schools 5909 Paris Road Columbia, MO 65202

Dear Randy Jackson:

Please find enclosed the analytical results for the sample(s) the laboratory received on **6/7/16 8:00 am** and logged in under work order **6061901**. All testing is performed according to our current TNI certifications unless otherwise noted. This report cannot be reproduced, except in full, without the written permission of PDC Laboratories, Inc.

If you have any questions regarding your report, please contact your project manager. Quality and timely data is of the utmost importance to us.

PDC Laboratories, Inc. appreciates the opportunity to provide you with analytical expertise. We are always trying to improve our customer service and we welcome you to contact the Vice President, John LaPayne with any feedback you have about your experience with our laboratory.

Sincerely,

Kurt Stepping Senior Project Manager

(309) 692-9688 x1719 kstepping@pdclab.com

TNI LABORATORI



PDC Laboratories, Inc.

2231 West Altorfer Drive Peoria, IL 61615 (800) 752-6651

ANALYTICAL RESULTS

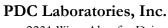
Sample: 6061901-01

Sampled: 05/31/16 00:00

-	6061901-01					•	05/31/16 (
Name: Alias:	WMS-KF HAND CENTER 1 KITCHEN SINK					Received: Matrix:	06/07/16 (งชะบบ Vater - Regular Sample
Alias.	TOTILIA OUAT						Dilliking v	vater - Regular Gample
arameter		Result	Unit	Qualifier	MCL	Analyzed	Analyst	Method
otal Metals - PIA	<u> </u>							
Copper		32	ug/L		1300	06/24/16 09:47	KMC	EPA 200.8
.ead		2.9	ug/L		15	06/24/16 09:47	KMC	EPA 200.8
Sample:	6061901-02					Sampled:	05/31/16 (00:00
Name:	WMS-KF KETTLE 2					Received:	06/07/16	08:00
Alias:	KITCHEN STEAM KETTLE					Matrix:	Drinking V	Vater - Regular Sample
Parameter		Result	Unit	Qualifier	MCL	Analyzed	Analyst	Method
otal Metals - PIA	4							
Copper		70	ug/L		1300	06/24/16 09:54	KMC	EPA 200.8
ead		1.3	ug/L		15	06/24/16 09:54	KMC	EPA 200.8
Sample:	6061901-03					Sampled:	05/31/16 (00:00
Name:	WMS-KF BAY SINK 3					Received:	06/07/16 (08:00
Alias:	KITCHEN 5 BAY SINK					Matrix:	Drinking V	Vater - Regular Sample
arameter		Result	Unit	Qualifier	MCL	Analyzed	Analyst	Method
otal Metals - PIA	<u> </u>							
Copper		20	ug/L		1300	06/24/16 10:00	KMC	EPA 200.8
ead		1.3	ug/L		15	06/24/16 10:00	KMC	EPA 200.8
Sample:	6061901-04					Sampled:	05/31/16 (00:00
Name:	WMS-FACSF 229 7						06/07/16 (
Alias:	RM 229 FIXTURE 7					Matrix:	Drinking V	Vater - Regular Sample

MCL Parameter Result Unit Qualifier Analyst Method Analyzed Total Metals - PIA Copper 1300 64 ug/L 06/24/16 10:01 KMC EPA 200.8 Lead 1.5 ug/L 15 06/24/16 10:01 KMC EPA 200.8

Customer #: 264641 www.pdclab.com Page 2 of 21





2231 West Altorfer Drive Peoria, IL 61615 (800) 752-6651

ANALYTICAL RESULTS

 Sample:
 6061901-05
 Sampled:
 05/31/16 00:00

 Name:
 WMS-FACSF 229 9
 Received:
 06/07/16 08:00

Alias: RM 229 FIXTURE 9 Matrix: Drinking Water - Regular Sample

Parameter Result Unit Qualifier MCL Analyzed Analyst Method Total Metals - PIA 1300 06/24/16 10:02 EPA 200.8 Copper 140 ug/L **KMC** Lead 1.5 15 06/24/16 10:02 **KMC** EPA 200.8 ug/L Sample: 6061901-06 Sampled: 05/31/16 00:00 Name: WMS-FACSF 229 10 Received: 06/07/16 08:00 Alias: RM 229 FIXTURE 11 Matrix: Drinking Water - Regular Sample

Analyst Method Result Unit Qualifier MCL **Parameter** Analyzed Total Metals - PIA 1300 Copper 120 ug/L 06/24/16 10:04 KMC EPA 200.8 Lead 3.3 15 06/24/16 10:04 KMC EPA 200.8 ug/L

 Sample:
 6061901-07
 Sampled:
 05/31/16 00:00

 Name:
 WMS-EWC HALL 13
 Received:
 06/07/16 08:00

Name: WWWS-EWC FIALL 13 Received: 00/07/10 00:00

Alias: OUTSIDE RM 227 #13 Matrix: Drinking Water - Regular Sample

Qualifier MCL **Parameter** Result Unit Analyzed Analyst Method Total Metals - PIA Copper 140 1300 06/24/16 10:05 KMC EPA 200.8 ug/L Lead 2.0 15 06/24/16 10:05 **KMC** EPA 200.8 ug/L

 Sample:
 6061901-08
 Sampled:
 05/31/16 00:00

 Name:
 WMS-TLF 221 15
 Received:
 06/07/16 08:00

Alias: RM 221 #15 Matrix: Drinking Water - Regular Sample

Qualifier MCL Result Unit Analyst Method **Parameter** Analyzed **Total Metals - PIA** Copper 91 ug/L 1300 06/24/16 10:06 **KMC** EPA 200.8 Lead 1.0 ug/L 15 06/24/16 10:06 KMC EPA 200.8

Customer #: 264641 www.pdclab.com Page 3 of 21



PDC Laboratories, Inc.

2231 West Altorfer Drive Peoria, IL 61615 (800) 752-6651

ANALYTICAL RESULTS

Sample: 6061901-09 Sampled: 05/31/16 00:00 WMS-EWC 222 16 Received: 06/07/16 08:00 Name:

OUTSIDE RM 222 #16

Alias: Matrix: Drinking Water - Regular Sample **Parameter** Result Unit Qualifier MCL Analyzed Analyst Method Total Metals - PIA 170 1300 06/24/16 10:07 EPA 200.8 Copper ug/L **KMC** Lead 1.6 15 06/24/16 10:07 **KMC** EPA 200.8 ug/L Sample: 6061901-10 Sampled: 05/31/16 00:00 Name: WMS-NSF 222 17 Received: 06/07/16 08:00 Alias: RM 222 OFFICE J #17 Matrix: Drinking Water - Regular Sample Analyst Method Result Unit Qualifier MCL **Parameter** Analyzed Total Metals - PIA

1300 Copper 120 ug/L 06/24/16 10:14 KMC EPA 200.8 Lead 3.9 15 06/24/16 10:14 KMC EPA 200.8 ug/L

Sample: 6061901-11 05/31/16 00:00 Sampled: Name: WMS-EWC 212 18 Received: 06/07/16 08:00

OUTSIDE RM 212 #18 Drinking Water - Regular Sample Alias: Matrix:

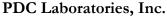
Qualifier MCL **Parameter** Result Unit Analyzed Analyst Method Total Metals - PIA Copper 130 1300 06/24/16 10:20 KMC EPA 200.8 ug/L Lead 1.8 15 06/24/16 10:20 **KMC** EPA 200.8 ug/L

Sampled: 05/31/16 00:00 Sample: 6061901-12 Name: WMS-EWC 211 19 Received: 06/07/16 08:00

OUTSIDE RM 211 #19 Matrix: Drinking Water - Regular Sample Alias:

Qualifier MCL Result Unit Analyst Method **Parameter** Analyzed **Total Metals - PIA** Copper 43 ug/L 1300 06/24/16 10:22 **KMC** EPA 200.8 Lead 8.5 ug/L 15 06/24/16 10:22 KMC EPA 200.8

Page 4 of 21 Customer #: 264641 www.pdclab.com





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ANALYTICAL RESULTS

 Sample:
 6061901-13
 Sampled:
 05/31/16 00:00

 Name:
 WMS-EWC 107 20
 Received:
 06/07/16 08:00

Name: **OUTSIDE RM 107 #20** Alias: Matrix: Drinking Water - Regular Sample **Parameter** Result Unit Qualifier MCL Analyzed Analyst Method Total Metals - PIA 1300 EPA 200.8 Copper 110 ug/L 06/24/16 10:23 **KMC** Lead 1.2 15 06/24/16 10:23 **KMC** EPA 200.8 ug/L Sample: 6061901-14 Sampled: 05/31/16 00:00 Name: WMS-EWC 112 21 Received: 06/07/16 08:00 Alias: **OUTSIDE RM 112 #21** Matrix: Drinking Water - Regular Sample Analyst Method Result Unit Qualifier MCL **Parameter** Analyzed Total Metals - PIA 1300 Copper 220 ug/L 06/24/16 10:24 KMC EPA 200.8 Lead 1.5 15 06/24/16 10:24 KMC EPA 200.8 ug/L

 Sample:
 6061901-15
 Sampled:
 05/31/16 00:00

 Name:
 WMS-EWC 123 22
 Received:
 06/07/16 08:00

Alias: OUTSIDE RM 123 #22 Matrix: Drinking Water - Regular Sample

Parameter Result Unit Qualifier MCL Analyzed Analyst Method

 Total Metals - PIA

 Copper
 160
 ug/L
 1300
 06/24/16 10:25
 KMC
 EPA 200.8

 Lead
 2.5
 ug/L
 15
 06/24/16 10:25
 KMC
 EPA 200.8

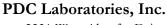
 Sample:
 6061901-16
 Sampled:
 05/31/16 00:00

 Name:
 WMS-EWC 101 23
 Received:
 06/07/16 08:00

Alias: OUTSIDE RM 101 #23 Matrix: Drinking Water - Regular Sample

Qualifier MCL Result Unit **Parameter** Analyzed Analyst Method **Total Metals - PIA** Copper 130 ug/L 1300 06/24/16 10:27 **KMC** EPA 200.8 Lead 7.2 ug/L 15 06/24/16 10:27 KMC EPA 200.8

Customer #: 264641 www.pdclab.com Page 5 of 21





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ANALYTICAL RESULTS

Sampled: 05/31/16 00:00 Sample: 6061901-17 WMS-EWC 136 24 Received: 06/07/16 08:00 Name:

OUTSIDE RM 136 #24 Alias: Matrix: Drinking Water - Regular Sample

Parameter Result Unit Qualifier MCL Analyzed Analyst Method Total Metals - PIA 91 1300 Copper ug/L 06/24/16 10:28 **KMC** EPA 200.8 Lead 3.4 15 06/24/16 10:28 **KMC** EPA 200.8 ug/L

Sample: 6061901-18 Sampled: 05/31/16 00:00

Name: WMS-EWC 136L 25 Received: 06/07/16 08:00

Alias: GIRL LOCKERRM 136L #25 Matrix: Drinking Water - Regular Sample

Analyst Method Result Unit Qualifier MCL **Parameter** Analyzed Total Metals - PIA 1300 Copper 92 ug/L 06/24/16 10:35 KMC EPA 200.8 Lead 1.4 15 06/24/16 10:35 KMC EPA 200.8 ug/L

Sample: 6061901-19 05/31/16 00:00 Sampled:

Unit

Name: WMS-EWC 137L 26 Received: 06/07/16 08:00

BOYS LOCKERRM 137L #26 Drinking Water - Regular Sample Alias: Matrix:

MCL

Qualifier **Parameter** Result Analyzed Analyst Method Total Metals - PIA Copper 150 1300 06/24/16 10:41 KMC EPA 200.8 ug/L Lead < 1.0 15 06/24/16 10:41 **KMC** EPA 200.8 ug/L

Sampled: 05/31/16 00:00 Sample: 6061901-20 Name:

WMS-EWC 135A 27 Received: 06/07/16 08:00

WOODSHOP #135A #27 Matrix: Drinking Water - Regular Sample Alias:

Qualifier MCL Result Unit **Parameter** Analyzed Analyst Method **Total Metals - PIA** Copper 180 ug/L 1300 06/24/16 10:42 **KMC** EPA 200.8 Lead 10 ug/L 15 06/24/16 10:42 KMC EPA 200.8

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Lead

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ANALYTICAL RESULTS

Sample: 6061901-21

Sampled: 05/31/16 00:00

Name:							06/07/16	
Alias:	OUTSIDE RM 1250 #29					Matrix:	Drinking V	Vater - Regular Sample
Parameter		Result	Unit	Qualifier	MCL	Analyzed	Analyst	Method
<u>Total Metals - F</u>	<u>PIA</u>							
Copper		100	ug/L		1300	06/24/16 10:43	KMC	EPA 200.8
Lead		< 1.0	ug/L		15	06/24/16 10:43	KMC	EPA 200.8
Sample	e: 6061901-22					Sampled:	05/31/16 (00:00
Name:							06/07/16 (
Alias:	OUTSIDE RM 1300 #30					Matrix:	Drinking V	Vater - Regular Sample
Parameter		Result	Unit	Qualifier	MCL	Analyzed	Analyst	Method
<u>Total Metals - F</u>	<u>PIA</u>							
Copper		180	ug/L		1300	06/24/16 10:45	KMC	EPA 200.8
Lead		< 1.0	ug/L		15	06/24/16 10:45	KMC	EPA 200.8
Sample	e: 6061901-23					Sampled:	05/31/16 (00:00
Name:	WMS-EWC BAND 33					Received:	06/07/16 (
Alias:	BAND RM 132 #33					Matrix:	Drinking V	Vater - Regular Sample
Parameter		Result	Unit	Qualifier	MCL	Analyzed	Analyst	Method
<u>Total Metals - F</u>	PIA .							
Copper		130	ug/L		1300	06/24/16 10:46	KMC	EPA 200.8
Lead		< 1.0	ug/L		15	06/24/16 10:46	KMC	EPA 200.8
Sample	e: 6061901-24					Sampled:	05/31/16 (00:00
Name:	WMS-EWC CHOIR 35					Received:	06/07/16 (08:00
Alias:	CHOIR RM 133 #35					Matrix:	Drinking V	Vater - Regular Sample
Parameter		Result	Unit	Qualifier	MCL	Analyzed	Analyst	Method
<u>Total Metals - F</u>	PIA .							
Copper		100	ug/L		1300	06/24/16 10:47	KMC	EPA 200.8

Page 7 of 21 Customer #: 264641 www.pdclab.com

15

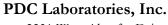
06/24/16 10:47

KMC

EPA 200.8

< 1.0

ug/L





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ANALYTICAL RESULTS

Sample: 6061901-25 Sampled: 06/03/16 00:00 WMS-131 CF 1 Received: 06/07/16 08:00 Name: **CLASSROOM COLD** Alias: Matrix: Drinking Water - Regular Sample **Parameter** Result Unit Qualifier MCL Analyzed Analyst Method Total Metals - PIA 100 1300 EPA 200.8 Copper ug/L 06/24/16 10:49 **KMC** Lead 2.9 15 06/24/16 10:49 **KMC** EPA 200.8 ug/L Sample: 6061901-26 Sampled: 06/03/16 00:00 Name: WMS-131 CF 2 Received: 06/07/16 08:00 Alias: **CLASSROOM COLD** Matrix: Drinking Water - Regular Sample Analyst Method Result Unit Qualifier MCL **Parameter** Analyzed Total Metals - PIA 220 Copper ug/L 1300 06/24/16 10:55 KMC EPA 200.8 Lead 16 **15 06/24/16 10:55 KMC EPA 200.8 ug/L 06/03/16 00:00 Sample: 6061901-27 Sampled: Name: WMS-131 CF 3 Received: 06/07/16 08:00 **CLASSROOM COLD** Drinking Water - Regular Sample Alias: Matrix: MCL Qualifier **Parameter** Result Unit Analyzed Analyst Method Total Metals - PIA Copper 100 1300 06/24/16 11:01 KMC EPA 200.8 ug/L

 Copper
 100
 ug/L
 1300
 06/24/16 11:01
 KMC
 EPA 200.8

 Lead
 2.8
 ug/L
 15
 06/24/16 11:01
 KMC
 EPA 200.8

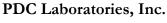
 Sample:
 6061901-28
 Sampled:
 06/03/16 00:00

 Name:
 WMS-124 CF 4
 Received:
 06/07/16 08:00

Alias: CLASSROOM COLD Matrix: Drinking Water - Regular Sample

Qualifier MCL Result Unit **Parameter** Analyzed Analyst Method **Total Metals - PIA** Copper 91 ug/L 1300 06/24/16 11:03 **KMC** EPA 200.8 Lead 1.5 ug/L 15 06/24/16 11:03 KMC EPA 200.8

Customer #: 264641 www.pdclab.com





Parameter

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ANALYTICAL RESULTS

 Sample:
 6061901-29
 Sampled:
 06/03/16 00:00

 Name:
 WMS-124 CF 5
 Received:
 06/07/16 08:00

Unit

Result

Alias: CLASSROOM COLD Matrix: Drinking Water - Regular Sample

Qualifier

MCL

Analyzed

Analyst Method

 Total Metals - PIA

 Copper
 92
 ug/L
 1300
 06/24/16 11:04
 KMC
 EPA 200.8

 Lead
 3.0
 ug/L
 15
 06/24/16 11:04
 KMC
 EPA 200.8

 Sample:
 6061901-30
 Sampled:
 06/03/16 00:00

 Name:
 WMS-124 CF 6
 Received:
 06/07/16 08:00

Alias: CLASSROOM COLD Matrix: Drinking Water - Regular Sample

Analyst Method Result Unit Qualifier MCL **Parameter** Analyzed Total Metals - PIA 1300 Copper 130 ug/L 06/24/16 11:05 KMC EPA 200.8 Lead 2.4 15 06/24/16 11:05 KMC EPA 200.8 ug/L

 Sample:
 6061901-31
 Sampled:
 06/03/16 00:00

 Name:
 WMS-1250 CF 7
 Received:
 06/07/16 08:00

Alias: CLASSROOM COLD Matrix: Drinking Water - Regular Sample

MCL Qualifier **Parameter** Result Unit Analyzed Analyst Method Total Metals - PIA Copper 130 1300 06/24/16 11:07 KMC EPA 200.8 ug/L Lead 2.8 15 06/24/16 11:07 **KMC** EPA 200.8 ug/L

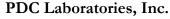
 Sample:
 6061901-32
 Sampled:
 06/03/16 00:00

 Name:
 WMS-125 CF 8
 Received:
 06/07/16 08:00

Alias: CLASSROOM COLD Matrix: Drinking Water - Regular Sample

Qualifier MCL Result Unit **Parameter** Analyzed Analyst Method **Total Metals - PIA** Copper 120 ug/L 1300 06/24/16 11:08 **KMC** EPA 200.8 Lead 2.4 ug/L 15 06/24/16 11:08 KMC EPA 200.8

Customer #: 264641 www.pdclab.com Page 9 of 21



Sampled: 06/03/16 00:00

Received: 06/07/16 08:00



Name:

Sample: 6061901-33

WMS-125 CF 9

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ANALYTICAL RESULTS

CLASSROOM COLD Alias: Matrix: Drinking Water - Regular Sample **Parameter** Result Unit Qualifier MCL Analyzed Analyst Method Total Metals - PIA 100 1300 06/24/16 11:09 EPA 200.8 Copper ug/L **KMC** Lead 1.3 15 06/24/16 11:09 **KMC** EPA 200.8 ug/L Sample: 6061901-34 Sampled: 06/03/16 00:00 Name: WMS-126 CF 10 Received: 06/07/16 08:00 Alias: **CLASSROOM COLD** Matrix: Drinking Water - Regular Sample Analyst Method Result Unit Qualifier MCL **Parameter** Analyzed Total Metals - PIA 1300 Copper 130 ug/L 06/24/16 11:16 KMC EPA 200.8 Lead 1.4 15 06/24/16 11:16 KMC EPA 200.8 ug/L

06/03/16 00:00 Sample: 6061901-35 Sampled: Name: WMS-126 CF 11 Received: 06/07/16 08:00 **CLASSROOM COLD**

Drinking Water - Regular Sample Alias: Matrix:

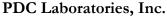
MCL Qualifier **Parameter** Result Unit Analyzed Analyst Method Total Metals - PIA Copper 100 1300 06/24/16 11:22 KMC EPA 200.8 ug/L Lead 1.5 15 06/24/16 11:22 **KMC** EPA 200.8 ug/L

Sampled: 06/03/16 00:00 Sample: 6061901-36 Name: WMS-126 CF 12 Received: 06/07/16 08:00 Alias:

CLASSROOM COLD Matrix: Drinking Water - Regular Sample

Qualifier MCL Result Unit **Parameter** Analyzed Analyst Method **Total Metals - PIA** Copper 130 ug/L 1300 06/24/16 11:23 **KMC** EPA 200.8 Lead 3.5 ug/L 15 06/24/16 11:23 KMC EPA 200.8

Customer #: 264641 www.pdclab.com





Parameter

Copper

Lead

Total Metals - PIA

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ANALYTICAL RESULTS

Sampled: 06/03/16 00:00 Sample: 6061901-37 WMS-1260 CF 13 Received: 06/07/16 08:00 Name: **CLASSROOM COLD** Alias: Matrix: Drinking Water - Regular Sample **Parameter** Result Unit Qualifier MCL Analyzed Analyst Method Total Metals - PIA 100 1300 06/24/16 11:25 EPA 200.8 Copper ug/L **KMC** Lead 1.3 15 06/24/16 11:25 **KMC** EPA 200.8 ug/L Sample: 6061901-38 Sampled: 06/03/16 00:00 Name: WMS-127 CF 14 Received: 06/07/16 08:00 Alias: **CLASSROOM COLD** Matrix: Drinking Water - Regular Sample Analyst Method Result Unit Qualifier MCL **Parameter** Analyzed Total Metals - PIA 1300 Copper 140 ug/L 06/24/16 11:26 KMC EPA 200.8 Lead 3.2 15 06/24/16 11:26 KMC EPA 200.8 ug/L 06/03/16 00:00 Sample: 6061901-39 Sampled: Name: WMS-127 CF 15 Received: 06/07/16 08:00 **CLASSROOM COLD** Drinking Water - Regular Sample Alias: Matrix: MCL Qualifier **Parameter** Result Unit Analyzed Analyst Method Total Metals - PIA Copper 79 1300 06/24/16 11:27 KMC EPA 200.8 ug/L Lead 1.3 15 06/24/16 11:27 **KMC** EPA 200.8 ug/L Sampled: 06/03/16 00:00 Sample: 6061901-40 Name: WMS-128 CF 16 Received: 06/07/16 08:00 **CLASSROOM COLD** Matrix: Drinking Water - Regular Sample Alias:

Customer #: 264641 www.pdclab.com Page 11 of 21

Qualifier

Result

110

< 1.0

Unit

ug/L

ug/L

MCL

1300

15

Analyzed

06/24/16 11:28

06/24/16 11:28

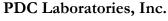
Analyst Method

EPA 200.8

EPA 200.8

KMC

KMC



Sampled: 06/03/16 00:00



Sample: 6061901-41

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ANALYTICAL RESULTS

WMS-128 CF 17 Received: 06/07/16 08:00 Name: **CLASSROOM COLD** Alias: Matrix: Drinking Water - Regular Sample **Parameter** Result Unit Qualifier MCL Analyzed Analyst Method Total Metals - PIA 1300 EPA 200.8 Copper 110 ug/L 06/24/16 11:30 **KMC** Lead 2.7 15 06/24/16 11:30 **KMC** EPA 200.8 ug/L Sample: 6061901-42 Sampled: 06/03/16 00:00 Name: WMS-128 CF 18 Received: 06/07/16 08:00 Alias: **CLASSROOM COLD** Matrix: Drinking Water - Regular Sample Analyst Method Result Unit Qualifier MCL **Parameter** Analyzed Total Metals - PIA 1300 Copper 120 ug/L 06/24/16 11:36 KMC EPA 200.8 Lead 2.5 15 06/24/16 11:36 KMC EPA 200.8 ug/L 06/03/16 00:00

 Sample:
 6061901-43
 Sampled:
 06/03/16 00:00

 Name:
 WMS-1280 CF 19
 Received:
 06/07/16 08:00

Alias: CLASSROOM COLD Matrix: Drinking Water - Regular Sample

MCL Qualifier **Parameter** Result Unit Analyzed Analyst Method Total Metals - PIA Copper 120 1300 06/24/16 11:43 KMC EPA 200.8 ug/L Lead 2.9 15 06/24/16 11:43 **KMC** EPA 200.8 ug/L

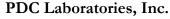
 Sample:
 6061901-44
 Sampled:
 06/03/16 00:00

 Name:
 WMS-129 CF 20
 Received:
 06/07/16 08:00

Alias: CLASSROOM COLD Matrix: Drinking Water - Regular Sample

Qualifier MCL Result Unit Analyst Method **Parameter** Analyzed **Total Metals - PIA** Copper 98 ug/L 1300 06/24/16 11:44 **KMC** EPA 200.8 Lead 2.1 ug/L 15 06/24/16 11:44 KMC EPA 200.8

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CLASSROOM COLD

Alias:

Parameter

Copper

Lead

Total Metals - PIA

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ANALYTICAL RESULTS

Sample: 6061901-45 Sampled: 06/03/16 00:00 WMS-129 CF 21 Received: 06/07/16 08:00 Name: **CLASSROOM COLD** Alias: Matrix: Drinking Water - Regular Sample **Parameter** Result Unit Qualifier MCL Analyzed Analyst Method Total Metals - PIA 1300 06/24/16 11:45 EPA 200.8 Copper 120 ug/L **KMC** Lead 1.8 15 06/24/16 11:45 **KMC** EPA 200.8 ug/L Sample: 6061901-46 Sampled: 06/03/16 00:00 Name: WMS-129 CF 22 Received: 06/07/16 08:00 Alias: **CLASSROOM COLD** Matrix: Drinking Water - Regular Sample Analyst Method Result Unit Qualifier MCL **Parameter** Analyzed Total Metals - PIA 1300 Copper 110 ug/L 06/24/16 11:46 KMC EPA 200.8 Lead 1.6 15 06/24/16 11:46 KMC EPA 200.8 ug/L 06/03/16 00:00 Sample: 6061901-47 Sampled: Name: WMS-130 CF 23 Received: 06/07/16 08:00 **CLASSROOM COLD** Drinking Water - Regular Sample Alias: Matrix: MCL Qualifier **Parameter** Result Unit Analyzed Analyst Method Total Metals - PIA Copper 130 1300 06/24/16 11:48 KMC EPA 200.8 ug/L Lead 3.6 15 06/24/16 11:48 **KMC** EPA 200.8 ug/L Sampled: 06/03/16 00:00 Sample: 6061901-48 Name: WMS-130 CF 24 Received: 06/07/16 08:00

Customer #: 264641 www.pdclab.com Page 13 of 21

Qualifier

Result

130

9.8

Unit

ug/L

ug/L

MCL

1300

15

Matrix:

Analyzed

06/24/16 11:49

06/24/16 11:49

Drinking Water - Regular Sample

EPA 200.8

EPA 200.8

Analyst Method

KMC

KMC



Parameter

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ANALYTICAL RESULTS

 Sample:
 6061901-49
 Sampled:
 06/03/16 00:00

 Name:
 WMS-130 CF 25
 Received:
 06/07/16 08:00

Unit

Result

Alias: CLASSROOM COLD Matrix: Drinking Water - Regular Sample

Qualifier

MCL

Analyzed

Analyst Method

 Total Metals - PIA

 Copper
 130
 ug/L
 1300
 06/24/16 11:50
 KMC
 EPA 200.8

 Lead
 3.3
 ug/L
 15
 06/24/16 11:50
 KMC
 EPA 200.8

 Sample:
 6061901-50
 Sampled:
 06/03/16 00:00

 Name:
 WMS-1300 CF 26
 Received:
 06/07/16 08:00

Alias: CLASSROOM COLD Matrix: Drinking Water - Regular Sample

Analyst Method Result Unit Qualifier MCL **Parameter** Analyzed Total Metals - PIA 1300 Copper 120 ug/L 06/24/16 12:00 KMC EPA 200.8 Lead 2.2 15 06/24/16 12:00 KMC EPA 200.8 ug/L

 Sample:
 6061901-51
 Sampled:
 06/03/16 00:00

 Name:
 WMS-111 CF 27
 Received:
 06/07/16 08:00

Alias: CLASSROOM COLD Matrix: Drinking Water - Regular Sample

MCL Qualifier **Parameter** Result Unit Analyzed Analyst Method Total Metals - PIA Copper 110 1300 06/24/16 12:06 KMC EPA 200.8 ug/L Lead 5.9 15 06/24/16 12:06 **KMC** EPA 200.8 ug/L

 Sample:
 6061901-52
 Sampled:
 06/03/16 00:00

 Name:
 WMS-1235 CF 28
 Received:
 06/07/16 08:00

Alias: CLASSROOM COLD Matrix: Drinking Water - Regular Sample

Qualifier MCL Result Unit **Parameter** Analyzed Analyst Method **Total Metals - PIA** Copper 110 ug/L 1300 06/24/16 12:08 **KMC** EPA 200.8 Lead 440 ug/L **15 06/24/16 12:08 KMC EPA 200.8

Customer #: 264641 www.pdclab.com



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ANALYTICAL RESULTS

Sample: 6061901-53 Name: WMS-123 CF 29

WWS-123 CF 29

Alias: CLASSROOM COLD

Sampled: 06/03/16 00:00

Received: 06/07/16 08:00

Matrix: Drinking Water - Regular Sample

Parameter	Result	Unit	Qualifier	MCL	Analyzed	Analyst	Method
Total Metals - PIA							
Copper	180	ug/L		1300	06/24/16 12:09	KMC	EPA 200.8
Lead	73	ug/L		**15	06/24/16 12:09	KMC	EPA 200.8

pcc

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NOTES

Specific method revisions used for analysis are available upon request.

** Indicates lab result exceeds a monitoring limit. Monitoring limits are either client permit limits or client requested action levels.

Certifications

PIA - Peoria, IL

TNI Accreditation for Drinking Water, Wastewater, Hazardous and Solid Wastes Fields of Testing through IL EPA Lab No. 100230 Illinois Department of Public Health Bacteriological Analysis in Drinking Water Approved Laboratory Registry No. 17553 Missouri Department of Natural Resources Certificate of Approval for Microbiological Laboratory Service No. 870 Drinking Water Certifications: Iowa (240); Kansas (E-10338); Missouri (870) Wastewater Certifications: Arkansas (88-0677); Iowa (240); Kansas (E-10338) Hazardous/Solid Waste Certifications: Arkansas (88-0677); Iowa (240); Kansas (E-10338)

SPMO - Springfield, MO USEPA DMR-QA Program

STL - St. Louis, MO

TNI Accreditation for Wastewater, Hazardous and Solid Wastes Fields of Testing through KS Lab No. E-10389 Illinois Department of Public Health Bacteriological Analysis in Drinking Water Approved Laboratory Registry No. 171050 Drinking Water Certifications: Missouri (1050) Missouri Department of Natural Resources

* Not a TNI accredited analyte

Just Sta

TNI TROORATORY

Certified by: Kurt Stepping, Senior Project Manager

Customer #: 264641 www.pdclab.com Page 16 of 21

West widdle School

Please Note: A maximum of twelve (12) samples per box.



Š.	Coding system	Description
2	WMS-EWC-107-20	Ewic outside Room 107 number 20
Ž	WMS- Euc-112-21	Ewc outside boom 112 number 21
15	WMS-Fuc-123-22	
16	WMS-EWC-101-23	Ewc outside Noom 101 number 23
17	WMS- EWC-136-24	outside Noom 136
18	WMS-6wc-1366- 25	Ewe in girls locker loom 1362 number 25
61	WMS-EWC-1376-26	EWC in Boys locker Room 137L number 26
07	WMS-Ewc - 135m- 27	EWC in wood shop loom #135A 11 contager 27
17	WMS- £WC- 1250-29	Ewc out side Noom 125 & number 29
22	WMS- Ewc - 1346-30	num ber
23	WMS- Ewc-Bard-33	132
24	WMS-Ewc-choir 35	inchoir Room 133 nu
Chai	Chain of Custody Section	
Sam	Sample Collector: (Print)	See Pase 1 (Signature) 7 (
Reli	Relinquished by: Kan	Date/Time: 6. 6 Received by: 100 100 Date/Time: 5-31-
Reli	Relinquished by:	Received by: What has been bate/Time: 6
Relii	Relinquished by:	Date/Time: 6716 Received by/funded(W) Date/Time: 6/7/(68/90)
	7	Sec.

Please Note: A maximum of twelve (12) samples per box.



														Date/Time: C/1//6 76 /6 us. Date/Time: C/1//6 /6 /22 Date/Time: C/1//6 &: O
	Class room Cold	, , , , , , , , , , , , , , , , , , , ,	1. 11	11 11	71 11	" "	71	,	F. 1.	2.2		7.	1 1 1	Received by: June Holy Received by: June Holy Received by: June Holy Received by: June Holy Holy Holy Holy Holy Holy Holy Holy
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No. Coding system	25 WMS- 131-CF-1	26 WMS-131-CF. 2	27 WMS- 131-CF. 3	78 WMS-124-Ct-4	27 WMS-124-CF-5	30 WMS- 124-CF-6	31 WMS- 1250-CE-7	32 WMS- 125-CF-8	33 WMS- 125-CF-9	34 WMS- 126-CF-10	35 WMS- 126-CF.(1	36 WMS- 126-CF. 12	Chain of Custody Section	Sample Collector: (Print) rav Relinquished by: Colton Davidon Relinquished by: Model

Please Note: A maximum of twelve (12) samples per box.



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Please Note: A maximum of twelve (12) samples per box.



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tion	de Classmon Gold		, ,	11 11	77								Smith (Signature) Fair Luna Date/Time: 6/8 Received by: 20/1/1/1/1/1/1/1/1/1/1/1/1/1/1/1/1/1/1/1
No. Coding system Description	7.	50 WMS- 1300-CF-26	51 WMS-111 - CF27	52 WMS-1235-CF-28	53 WMS-123-CF-29	WMS-	Chain of Custody Section Sample Collector: (Print) Traus Smith Relinquished by: Lotton Braidson Date/Time: 6/2 Relinquished by: Act Time: 6/2 Relinquished by: Act Time: 6/2						

Please Note: A maximum of twelve (12) samples per box.

Facilities & Construction Services

PDC Laboratories Inc. Laboratory Results Follow-up Water Quality Evaluation

Field Data & Chain-of-Custody Form





PDC Laboratories, Inc.

PROFESSIONAL • DEPENDABLE • COMMITTED

July 18, 2016

Randy Jackson Columbia Public Schools 5909 Paris Road Columbia, MO 65202

Dear Randy Jackson:

Please find enclosed the analytical results for the sample(s) the laboratory received on **7/11/16 4:00 pm** and logged in under work order **6071529**. All testing is performed according to our current TNI certifications unless otherwise noted. This report cannot be reproduced, except in full, without the written permission of PDC Laboratories, Inc.

If you have any questions regarding your report, please contact your project manager. Quality and timely data is of the utmost importance to us.

PDC Laboratories, Inc. appreciates the opportunity to provide you with analytical expertise. We are always trying to improve our customer service and we welcome you to contact the Vice President, John LaPayne with any feedback you have about your experience with our laboratory.

Sincerely,

Kurt Stepping

Senior Project Manager (309) 692-9688 x1719 kstepping@pdclab.com





PDC Laboratories, Inc.

2231 West Altorfer Drive Peoria, IL 61615 (800) 752-6651

ANALYTICAL RESULTS

Sample: 6071529-01 Name:

WMS-FS-131-CF-2

Alias:

CLASSROOM FAUCET (COLD) IN RM 131

Sampled: 07/07/16 00:00

Received: 07/11/16 16:00

Matrix: Drinking Water - Regular Sample

Parameter Result Unit Qualifier MCL Analyzed Analyst Method Total Metals - PIA Copper 85 1300 07/15/16 14:02 KMC EPA 200.8 ug/L Lead 1.2 15 07/15/16 14:02 KMC EPA 200.8 ug/L

Sample: 6071529-02

Name: WMS-FS-1235-CF-28

Alias: CLASSROOM FAUCET (COLD) IN RM 1235 Sampled: 07/07/16 00:00

Received: 07/11/16 16:00

Matrix: Drinking Water - Regular Sample

Parameter Result Unit Qualifier MCL Analyzed Analyst Method Total Metals - PIA 150 1300 07/15/16 14:06 Copper ug/L KMC EPA 200.8 Lead 18 **15 07/15/16 14:06 KMC EPA 200.8 ug/L

Sample: 6071529-03

Name: WMS-FS-123-CF-29

CLASSROOM FAUCET (COLD) IN RM 123 Alias:

Sampled: 07/07/16 00:00

Received: 07/11/16 16:00

Drinking Water - Regular Sample Matrix:

Qualifier MCL **Parameter** Result Unit Analyzed Analyst Method Total Metals - PIA Copper 82 ug/L 1300 07/15/16 14:07 KMC EPA 200.8 **15 Lead 22 07/15/16 14:07 KMC EPA 200.8 ug/L

Page 2 of 6 Customer #: 264641 www.pdclab.com

pcc

PDC Laboratories, Inc.

2231 West Altorfer Drive Peoria, IL 61615 (800) 752-6651

NOTES

Specific method revisions used for analysis are available upon request.

** Indicates lab result exceeds a monitoring limit. Monitoring limits are either client permit limits or client requested action levels.

Certifications

PIA - Peoria, IL

TNI Accreditation for Drinking Water, Wastewater, Hazardous and Solid Wastes Fields of Testing through IL EPA Lab No. 100230 Illinois Department of Public Health Bacteriological Analysis in Drinking Water Approved Laboratory Registry No. 17553 Missouri Department of Natural Resources Certificate of Approval for Microbiological Laboratory Service No. 870 Drinking Water Certifications: Iowa (240); Kansas (E-10338); Missouri (870) Wastewater Certifications: Arkansas (88-0677); Iowa (240); Kansas (E-10338) Hazardous/Solid Waste Certifications: Arkansas (88-0677); Iowa (240); Kansas (E-10338)

SPMO - Springfield, MO USEPA DMR-QA Program

STL - St. Louis, MO

TNI Accreditation for Wastewater, Hazardous and Solid Wastes Fields of Testing through KS Lab No. E-10389 Illinois Department of Public Health Bacteriological Analysis in Drinking Water Approved Laboratory Registry No. 171050 Drinking Water Certifications: Missouri (1050) Missouri Department of Natural Resources

* Not a TNI accredited analyte

Just Sta

TNI TNI TROORATORY

Certified by: Kurt Stepping, Senior Project Manager

Customer #: 264641 www.pdclab.com Page 3 of 6

SAMPLING METHOD

The water sample method was a "first-draw" sample. The first-draw sample was a sample collected first thing in the morning and is representative of the water that may be consumed at the beginning of the day or after infrequent use. Water samples were collected from cold water lines; the recommended procedure by the EPA - Reference: The Environmental Protection Agency (EPA) Lead and Copper Rule, 40 C.F.R. Sections 141 .80 to 141.91, Revised Version: February 2016

The following sample revealed an elevated lead and/or copper lab report:

Sample No.	Description/Location	Tag	CPS-FCS
26	Classroom Faucet (cold) in room 131	131-CF-2	Actions 1

Sample collected on June 3, 2016 and analyzed on June 24, 2016.

1 = CPS-FCS & PDC Laboratories recommend(s) the following actions:
(Part A) An inspection of the water tap and associated water line by a
CPS-FCS specialized maintenance plumber to help determine the cause,
(Part B) CPS-FCS shall conduct a follow-up sample collection-a flush
sample*, and (Part C) CPS-FCS shall post a "Notice" sign above the water
tap instructed the user to do not use this water source for consumable
purposes.

*Follow-up Sample(s)

This sample is representative of the water that is in the plumbing upstream from the faucet or drinking water fountain. This sample shall be collected before the facility opens and before any water was used at the water tap. The water from the faucet or drinking water fountain should run for 30 seconds before collecting the sample. This is called a flush sample.

Part A: CPS-FCS on-site review of: Classroom Faucet (cold) in room 131

Date: 7-7-16 Name of Reviewer: Chris Willows
Actions taken: Inspected faucet
Comments: Cold handle is leaking and needs repair. This is
a lat sink and should not be used for consumption.
Signature of Reviewer: (Mrs. M/hz.

601/529-3 MJ

West Middle School

(Part B) CPS-FCS shall conduct a follow-up sample collection-a flush sample*,

Sample	Description/Loc	cation		Tag				
No.	_							
1	Classroom Faucet	(cold) in room	131	FS-131-CF-2				
Chain of Custody Section								
Sample Co	ollector (Print):	60		Signature:				
Relinguish	led by:	Date/Time:	- NOON	Received by	Date/Time: 7-11-16 1 200			
Relinguish	ad by:	Bate/Time:	1600	Received by:	Date/Time:			
Rélinquish	ned by:	Date/Time:	4 <i>V</i> /	Received by	Date/Time: 7/11/6 (600			

EWC Electronic Water Cooler (chiller unit)

DWF Drinking Water Faucet (bubbler/fountain)

CF Classroom Faucet (cold)
KF Kitchen Faucet (cold)

FACSF Family and Consumer Science Faucet (cold)

RRF Restroom Faucet (cold)
NSF Nurse's Sink Faucet (co

NSF Nurse's Sink Faucet (cold) TLF Teacher's Lounge Faucet

**FS Follow-up Sample

(Part C) CPS-FCS shall post a "Notice" sign above the water tap instructed the user to do not use this water source for consumable purposes.



SAMPLING METHOD

The water sample method was a "first-draw" sample. The first-draw sample was a sample collected first thing in the morning and is representative of the water that may be consumed at the beginning of the day or after infrequent use. Water samples were collected from cold water lines; the recommended procedure by the EPA - Reference: The Environmental Protection Agency (EPA) Lead and Copper Rule, 40 C.F.R. Sections 141 .80 to 141.91, Revised Version: February 2016

The following sample revealed an elevated lead and/or copper lab report:

Sample	Description/Location	Tag	CPS-FCS
No.			Actions
52	Classroom Faucet (cold) in room 1235	1235-CF-28	1

Sample collected on June 3, 2016 and analyzed on June 24, 2016.

1 = CPS-FCS & PDC Laboratories recommend(s) the following actions:

(Part A) An inspection of the water tap and associated water line by a CPS-FCS specialized maintenance plumber to help determine the cause, (Part B) CPS-FCS shall conduct a follow-up sample collection-a flush sample*, and (Part C) CPS-FCS shall post a "Notice" sign above the water tap instructed the user to do not use this water source for consumable purposes.

*Follow-up Sample(s)

This sample is representative of the water that is in the plumbing upstream from the faucet or drinking water fountain. This sample shall be collected before the facility opens and before any water was used at the water tap. The water from the faucet or drinking water fountain should run for 30 seconds before collecting the sample. This is called a flush sample.

Part A: CPS-FCS on-site review of: Classroom Faucet (cold) in room 1235

(Part B) CPS-FCS shall conduct a follow-up sample collection-a flush sample*,

Sample	Description/Lo	antina							
_	e Description/Loc	Jation	Tag						
No.									
2	Classroom Faucet	(cold) in room 1235	FS-1235-CF-28						
		Chain of Custody	Section	^					
Sample	Collector (Print):		Signature	7					
Dali	dy Jackson	T	A secon	100					
Relingui	shed by:	Date/Time: 7-11-16- NON	Received by:	Date/Time: 1100					
Relingu	Mas	Date/Time 1-11-16 1600	Received by:	Date/Time:					
Relinqui	shed by:	Date/Time:	Received by	Date/Time: 7/1-16/1600					
DILLO	D1	0 1 (1111							
EWC	Electronic Wat	er Cooler (chiller unit)							
DWF	Drinking Water	· Faucet (bubbler/fount	tain)						
CF	Classroom Fau								
KF	Kitchen Faucet	t (cold)							
FACSF	Family and Cor	nsumer Science Fauce	t (cold)						
RRF	Restroom Faucet (cold)								
NSF	Nurse's Sink Faucet (cold)								
TLF	Teacher's Loui	nge Faucet							
**FS	Follow-up Sam	ple							

(Part C) CPS-FCS shall post a "Notice" sign above the water tap instructed the user to do not use this water source for consumable purposes.



SAMPLING METHOD

The water sample method was a "first-draw" sample. The first-draw sample was a sample collected first thing in the morning and is representative of the water that may be consumed at the beginning of the day or after infrequent use. Water samples were collected from cold water lines; the recommended procedure by the EPA - Reference: The Environmental Protection Agency (EPA) Lead and Copper Rule, 40 C.F.R. Sections 141 .80 to 141.91, Revised Version: February 2016

The following sample revealed an elevated lead and/or copper lab report:

Sample No.	Description/Location	Tag	CPS-FCS Actions
53	Classroom Faucet (cold) in room 123	123-CF-29	1

Sample collected on June 3, 2016 and analyzed on June 24, 2016.

1 = CPS-FCS & PDC Laboratories recommend(s) the following actions: (Part A) An inspection of the water tap and associated water line by a CPS-FCS specialized maintenance plumber to help determine the cause, (Part B) CPS-FCS shall conduct a follow-up sample collection-a flush sample*, and (Part C) CPS-FCS shall post a "Notice" sign above the water tap instructed the user to do not use this water source for consumable

purposes.

*Follow-up Sample(s)

This sample is representative of the water that is in the plumbing upstream from the faucet or drinking water fountain. This sample shall be collected before the facility opens and before any water was used at the water tap. The water from the faucet or drinking water fountain should run for 30 seconds before collecting the sample. This is called a flush sample.

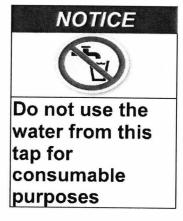
Part A: CPS-FCS on-site review of: Classroom Faucet (cold) in room 123

Date: 7-7-16 Name of Reviewer: (Wis Willows
Actions taken: Inspected faucet
Comments: This foucet is old and stitt but is working. This is
a lab sink and should be turned off if not in use.
Signature of Reviewer: Mrs Whan

(Part B) CPS-FCS shall conduct a follow-up sample collection-a flush sample*,

Sample	Description/Location	To						
No.	Bescription/Location	Tag						
	CI							
3	Classroom Faucet (cold) in room 123	FS-123-CF-29						
	Chain of	Custody Section						
Sample (Collector (Print):	Signature:						
para	> tockson	1 Dres Leton						
Relinqui	2000 de lon 7-11-16 - 1	Noon Redended by Date/Time:						
Relinqui	shell by: Date/Time:	Received by: Date/Time:						
Relinquis	shed by: Date/Time:	Received by Date/Time:						
EWC	Electronic Water Cooler (chille	er unit)						
DWF	Drinking Water Faucet (bubble	r/fountain)						
CF	Classroom Faucet (cold)	,						
KF	Kitchen Faucet (cold)							
FACSF	Family and Consumer Science	Faucet (cold)						
	Restroom Faucet (cold)							
NSF	Nurse's Sink Faucet (cold)							
TLF	Teacher's Lounge Faucet							
**FS	Follow-up Samples							

(Part C) CPS-FCS shall post a "Notice" sign above the water tap instructed the user to do not use this water source for consumable purposes.



Facilities & Construction Services

Appendix C



Facilities & Construction Services

PDC Laboratory Certifications



State of Missouri Department of Natural Resources

Certificate of Approval for Chemical Laboratory Service

This is to certify that

PDC Laboratories, Inc.

is hereby approved to perform the analysis of drinking water as specified on the Certified Parameter List, which must accompany this certificate to be valid.

Certification No.	870	- Holyn
Date Issued	September 16, 2015	Chief, Public Drinking Water Branch Water Protection Program Department of Natural Resources
Expiration Date	June 30, 2016	Director, Environmental Services Program Department of Natural Resources
		Chi Beldt
		Evaluation Officer Environmental Services Program

Department of Natural Resources

MISSOURI DEPARTMENT OF NATURAL RESOURCES

DRINKING WATER LABORATORY

CERTIFIED PARAMETER LIST

This is to certify that

PDC Laboratories, Inc.

located at

2231 West Altorfer Drive, Peoria, Illinois

has been approved to perform the indicated procedures on drinking water under the Missouri Public Drinking Water Regulations (10 CSR 60-5.020). Specific method numbers or references are included in parenthesis when appropriate.

METALS

EPA 200.7 - Aluminum, Arsenic, Barium, Beryllium, Cadmium, Calcium, Chromium, Copper, Hardness (calculated), Iron, Magnesium, Manganese, Nickel, Silica, Sodium, Silver, Zinc;
 EPA 200.8 - Aluminum, Antimony, Arsenic, Barium, Beryllium, Cadmium, Chromium, Copper, Lead, Manganese, Mercury, Molybdenum, Nickel, Selenium, Silver, Thallium, Zinc;
 EPA 245.1 - Mercury

INORGANIC NONMETALLIC CONSTITUENTS

SM 4500C-G - Chlorine (Free, Combined, Total);
 SM 4500F-C - Fluoride;
 SM 4500H-B - Hydrogen Ion (pH);
 SM 4500P-E - Orthophosphate;
 EPA 150.1 - Hydrogen Ion (pH);
 EPA 300.0 - Chloride,
 Fluoride,
 Nitrate,
 Nitrate,
 Fluoride;
 EPA 353.2 - Nitrate,
 Nitrite

PHYSICAL & AGGREGATE PROPERTIES

SM 2150B - Odor; SM 2320B - Alkalinity; SM 2330B - Corrosivity (Langlier Index);
 SM 2510B - Conductivity; SM 2540C - Total Dissolved Solids (TDS);
 EPA 180.1 - Turbidity

AGGREGATE ORGANIC CONSTITUENTS

SM 5310C – Total Organic Carbon (TOC); SM 5540C – Foaming Agents; SM 5910B – UV254

ORGANIC COMPOUNDS

EPA 1613RB – Dioxin (2,3,7,8 TCDD); *EPA 504.1* – 1,2-Dibromo-3-chloropropane (DBCP), 1,2-Dibromoethane (EDB); *EPA 515.3* – 2,4,5-TP (Silvex), 2,4-D, Dalapon, Dicamba, Dinoseb, Pentachlorophenol, Picloram;

EPA 524.2 – 1,1,1,2-Tetrachloroethane, 1,1,1-Trichloroethane, 1,1,2,2-Tetrachloroethane, 1,1,2-Trichloroethane, 1,1-Dichloroethane, 1,1-Dichloroethene, 1,1-Dichloropropene, 1,2,3-Trichlorobenzene, 1,2,3-Trichloropropane, 1,2,4-Trichlorobenzene, 1,2,4-Trimethylbenzene, 1,2-Dichlorobenzene, 1,2-Dichloroethane, 1,2-Dichloropropane, 1,3-Dichlorobenzene, 1,3-Dichloropropane, 1,3,5-Trimethylbenzene, 1,4-Dichlorobenzene, 2-Chlorotoluene, 2,2-Dichloropropane, 4-Chlorotoluene, 4-Isopropyltoluene, Benzene, Bromobenzene, Bromochloromethane, Bromodichloromethane, Bromoform, Bromomethane, Carbon Tetrachloride, Chlorobenzene, Chloroethane, Chloroform, Chloromethane, cis-1,2-Dichloroethene, cis-1,3-Dichloropropene, Dibromochloromethane, Dibromomethane, Dichlorodifluoromethane, Dichloromethane (Methylene Chloride), Ethylbenzene, Trichlorofluoromethane, Hexachlorobutadiene, Isopropylbenzene, Methyl tert-butyl ether (MTBE), Naphthalene, n-Butylbenzene, n-Propylbenzene, sec-Butylbenzene, Styrene, tert-Butylbenzene, Tetrachloroethene, Toluene, Total Trihalomethanes, trans-1,2-Dichloroethene, trans-1,3-Dichloropropene, Trichloroethylene (TCE), Vinyl Chloride, Xylenes (total); *EPA 525.2 – 4,4'-DDT*, Alachlor, Aldrin, Atrazine, Benzo(a)pyrene, Butachlor, Chlorodane (total), Di(2-ethylhexyl)adipate, Di(2-ethylhexl)phthalate, Dieldrin, Endrin, gamma-BHC (Lindane), Heptachlor, Heptachlor Epoxide, Hexachlorobenzene, Hexachlorocyclopentadiene, Methoxychlor, Metolachlor, Metribuzin, PCB as Aroclor, Propachlor, Simazine, Toxaphene; EPA 531.1 – 3-Hydroxycarbofuran, Aldicarb (Temik), Aldicarb Sulfone, Aldicarb Sulfoxide, Carbaryl (Sevin), Carbofuran (Furaden), Methomyl (Lannate), Oxamyl; *EPA 547* – Glyphosate; *EPA 548.1* – Endothall; *EPA 549.2* – Diquat; *EPA 550* – Beno(a)pyrene; EPA 552.2 – Dibromoacetic acid, Dichloroacetic acid, Monobromoacetic acid,

> Missouri Certificate No.: 870 Expiration Date: June 30, 2016 Original Certifying State: Illinois

Monochloroacetic acid, Trichloroacetic acid

Facilities & Construction Services

Appendix D



Facilities & Construction Services

Lead and Copper Rule



Date Signed: February 29, 2016



UNITED STATES ENVIRONMENTAL PROTECTION AGENCY WASHINGTON, D.C. 20460

FEB 2 9 2016

OFFICE OF WATER

MEMORANDUM

SUBJECT: Clarification of Recommended Tap Sampling Procedures for Purposes of the Lead and

Copper Rule

FROM: Peter C. Grevatt, Director

Office of Ground Water & Drinking Water

TO: Water Division Directors

Regions I - X

The Lead and Copper Rule, 40 C.F.R. Sections 141.80 to 141.91, requires monitoring at consumer taps to identify levels of lead in drinking water that may result from corrosion of lead-bearing components in a public water system's distribution system or in household plumbing. These samples help assess the need for, or the effectiveness of, corrosion control treatment. The purpose of this memorandum is to provide recommendations on how public water systems should address the removal and cleaning of aerators, pre-stagnation flushing, and bottle configuration for the purpose of Lead and Copper Rule sampling.

Removal and Cleaning of Aerators

EPA issued a memorandum on *Management of Aerators during Collection of Tap Samples to Comply with the Lead and Copper Rule* on October 20, 2006. This memorandum stated that EPA recommends that homeowners regularly clean their aerators to remove particulate matter as a general practice, but states that public water systems should not recommend the removal or cleaning of aerators prior to or during the collection of tap samples gathered for purposes of the Lead and Copper Rule. EPA continues to recommend this approach. The removal or cleaning of aerators during collection of tap samples could mask the added contribution of lead at the tap, which may potentially lead to the public water system not taking additional actions needed to reduce exposure to lead in drinking water. EPA's recommendation about the removal and cleaning of aerators during sample collection applies only to monitoring for lead and copper conducted pursuant to 40 C.F.R. 141.86.

Pre-Stagnation Flushing

EPA is aware that some sampling instructions provided to residents include recommendations to flush the tap for a specified period of time prior to starting the minimum 6-hour stagnation time required for samples collected under the Lead and Copper Rule. This practice is called pre-stagnation flushing. Pre-stagnation flushing may potentially lower the lead levels as compared to when it is not practiced.

Flushing removes water that may have been in contact with the lead service line for extended periods, which is when lead typically leaches into drinking water. Therefore, EPA recommends that sampling instructions not contain a pre-stagnation flushing step.

Bottle Configuration

EPA recommends that wide-mouth bottles be used to collect Lead and Copper compliance samples. It has become apparent that wide-mouth bottles offer advantages over narrow-necked bottles because wide-mouth bottles allow for a higher flow rate during sample collection which is more representative of the flow that a consumer may use to fill up a glass of water. In addition, a higher flow rate can result in greater release of particulate and colloidal lead and therefore is more conservative in terms of identifying lead concentrations.

Conclusion

EPA is providing these recommendations for collection of Lead and Copper Rule tap samples to better reflect the state of knowledge about the fate and transport of lead in distribution systems. The three areas discussed above may potentially lead to samples that erroneously reflect lower levels of lead concentrations. The recommendations in this memorandum are also consistent with the recommendations provided by the EPA's Flint Task Force. For more information about the Task Force please view EPA's website at: http://www.epa.gov/flint.

To provide further information on this topic, EPA included an amended "Suggested Directions for Homeowner Tap Sample Collection Procedures" in Appendix D of the 2010 revision of *Lead and Copper Rule Monitoring and Reporting Guidance for Public Water Systems* (EPA 816-R-10-004). This document can be found at:

http://nepis.epa.gov/Exe/ZyPDF.cgi?Dockey=P100DP2P.txt

Please share these recommendations with your state drinking water program directors. If you have any questions, please contact Anita Thompkins at thompkins.anita@epa.gov.

Attachment

cc: James Taft, Association of State Drinking Water Administrators

Suggested Directions for Homeowner Tap Sample Collection Procedures Revised Version: February 2016

These samples are being collected to determine the lead and copper levels in your tap water. This sampling effort is required by the U.S. Environmental Protection Agency and your State under the Lead and Copper Rule, and is being accomplished through a collaboration between the public water system and their consumers (e.g. residents).

Collect samples from a tap that has not been used for at least 6 hours. To ensure the water has not been used for at least 6 hours, the best time to collect samples is either early in the morning or in the evening upon returning from work. Be sure to use a kitchen or bathroom cold water tap that has been used for drinking water consumption in the past few weeks. The collection procedure is described below.

- Prior arrangements will be made with you, the customer, to coordinate the sample collection. Dates will be set for sample kit delivery and pick-up by water system staff.
- 2. There must be a minimum of 6 hours during which there is no water used from the tap where the sample will be collected and any taps adjacent or close to that tap. Either early mornings or evenings upon returning home are the best sampling times to ensure that the necessary stagnant water conditions exist. Do not intentionally flush the water line before the start of the 6 hour period.
- 3. Use a kitchen or bathroom cold-water faucet for sampling. If you have water softeners on your kitchen taps, collect your sample from the bathroom tap that is not attached to a water softener, or a point of use filter, if possible. Do not remove the aerator prior to sampling. Place the opened sample bottle below the faucet and open the cold water tap as you would do to fill a glass of water. Fill the sample bottle to the line marked "1000-mL" and turn off the water.
- Tightly cap the sample bottle and place in the sample kit provided. Please review the sample kit label at this time to ensure that all information contained on the label is correct.
- If any plumbing repairs or replacement has been done in the home since the previous sampling
 event, note this information on the label as provided. Also if your sample was collected from a tap
 with a water softener, note this as well.
- 6. Place the sample kit in the same location the kit was delivered to so that water system staff may pick up the sample kit.
- 7. Results from this monitoring effort and information about lead will be provided to you as soon as practical but no later than 30 days after the system learns of the tap monitoring results. However, if excessive lead and/or copper levels are found, immediate notification will be provided (usually 1-2 working days after the system learns of the tap monitoring results).

	TO BE CO	OMPLETED BY RESIDENT
Water was last used:	Time	Date
Sample was collected:	Time	Date
Sample Location & fau	ucet (e.g. Bathr	oom sink):

Facilities & Construction Services

Columbia Missouri Water and Light 2015 Water Testing Results



Columbia Missouri Water and Light: 2015 Water Testing Results WATER QUALITY CONTROL ANALYSIS OF TREATED WATER

QUALITY MEASURE	CONCENTRATION (milligrams per liter)
pН	8.33 (pH units)
Alkalinity (Total)*	123
Phenolphthalein Alkalinity*	14
Total Hardness*	168
Calcium (as calcium carbonate)*	100
Magnesium (as calcium carbonate)*	71
Sulfate**	86.3
Total Dissolved Solids**	309
Fluoride**	0.58
Free Chlorine	3.10
Total Chlorine	3.80
Chloride**	29.9
Iron**	0.12
Manganese**	0.02
Silica	21.4

^{*}State regulations require these tests to be run for operation information

<Means less than — Equipment used cannot measure the contaminant below this level.

CONTAMINANT	CONCENTRATION (milligrams per liter)	MAXIMUM CONTAMINANT LEVEL (milligrams per liter)
VOLATILE ORGANIC CHEMICA	LS:	
Trichloroethylene Tetrachloroethylene Carbon tetrachloride 1,1,1-trichloroethane	NONE DETECTED NONE DETECTED NONE DETECTED NONE DETECTED	0.005 0.005 0.005 0.20

1,2-dichloroethane	NONE DETECTED	0.005
Vinyl Chloride	NONE DETECTED	0.002
Benzene	NONE DETECTED	0.005
Dichloromethane	NONE DETECTED	0.005
Monochlorobenzene	NONE DETECTED	0.1
1,1-dichloroethylene	NONE DETECTED	0.007
-		

Cis-1,2-dichloroethylene	NONE DETECTED	0.7
Trans-1,2-dichloroethylene	NONE DETECTED	0.1

29.3

INORGANIC CHEMICALS:

Sodium**

Asbestos Arsenic Barium Cadmium Chromium	NONE DETECTED NONE DETECTED 0.123 NONE DETECTED 0.001	7 million fibers per liter 0.01 2 0.005 0.1
Lead Mercury Nitrate (as Nitrogen) Selenium Nitrite (as Nitrogen) Fluoride Aluminum** Antimony	0.0026 NONE DETECTED 0.03 NONE DETECTED NONE DETECTED 0.58 NONE DETECTED NONE DETECTED	Treatment technique, Action level = 0.015 0.002 10 0.05 1.0 4.0 0.05 0.006
Sulfate Copper**	86.3 0.040	250 Treatment technique, Action level = 1.3

No MCL, monitoring only

^{**}Secondary standards

CONTAMINANT	CONCENTRATION	MAXIMUM CONTAMINANT
	(milligrams per liter)	LEVEL (milligrams per liter)

INORGANIC CHEMICALS CONTINUED:

NONE DETECTED	0.1
1.95	5
NONE DETECTED	0.002
NONE DETECTED	0.004
NONE DETECTED	0.2
	1.95 NONE DETECTED NONE DETECTED

SYNTHETIC ORGANIC CHEMICALS:

TTHM	0.0499	0.080
Endrin	NONE DETECTED	0.002
Lindane	NONE DETECTED	0.002
Methoxychlor	NONE DETECTED	0.0002
Toxaphene	NONE DETECTED	0.003
Тохариене	NONE DETECTED	0.003
2,4-D	NONE DETECTED	0.07
2,4,5-TP (silvex)	NONE DETECTED	0.05
Aldicarb	NONE DETECTED	0.003
Aidicaio	NONE DETECTED	0.003
Chlordane	NONE DETECTED	0.002
Dalapon	NONE DETECTED	0.2
Diquat	NONE DETECTED	0.02
Endothall	NONE DETECTED	0.02
Endoman	NONE DETECTED	0.1
Glyphosate	NONE DETECTED	0.7
Carbofuran	NONE DETECTED	0.04
Vydate	NONE DETECTED	0.1
· y dute	TOTAL BETEGIEB	0.1
Simazine	NONE DETECTED	0.004
PAHs	NONE DETECTED	0.0002
PCBs	NONE DETECTED	0.0005
Atrazine	NONE DETECTED	0.003
Heptachlor	NONE DETECTED	0.0004
1		
Pentachlorophenol	NONE DETECTED	0.001
Picloram	NONE DETECTED	0.5
Dinoseb	NONE DETECTED	0.007
Alachlor	NONE DETECTED	0.002
Hexachlorocyclopentadiene	NONE DETECTED	0.05
2,3,7,8-TCDD	NONE DETECTED	0.00000003
HAA5	0.022	0.060
	0.022	0.000

MICROBIOLOGICAL CONTAMINANTS:

Total coliforms	NONE DETECTED	No more than 5% of Total Number of
		samples may be positive

RADIONUCLIDES: (Picocuries per Liter)

Radium	NONE DETECTED	5 picocuries per liter
Gross alpha	NONE DETECTED	15 picocuries per liter
Beta particle	6.6	4 mrem/yr 50 picocuries per liter
Uranium	0.48	20 ug/L
Radon	50.4	300

Facilities & Construction Services

EPA Drinking Water Best Management Practices



Drinking Water Best Management Practices

For Schools and Child Care Facilities Served by Municipal Water Systems



Drinking Water Best Management Practices

For Schools and Child Care Facilities Served by Municipal Water Systems

This guide is intended for use by school officials and child care providers responsible for the maintenance and/or safety of school and child care facilities including the drinking water. The purpose of this guide is to describe the importance of implementing best management practices for drinking water in schools and child care facilities and how a school or child care facility would go about implementing these practices. This guide is specifically for schools and child care facilities that receive water from water utilities or water suppliers such as cities, towns and water districts. This guide is not a regulation itself, nor does it change or substitute for those provisions and regulations. Thus, it does not impose legally binding requirements on EPA, states, municipal water systems, schools or child care facilities. This guide does not confer legal rights or impose legal obligations upon any member of the public. While EPA has made every effort to ensure the accuracy of the information in this guide the obligations of the regulated community are determined by statutes, regulations or other legally binding requirements. In the event of a conflict between the information in this guide and any statute or regulation, this document would not be controlling.

Table of Contents

What Decision Makers Should Know	1
Bacteria	2
Lead	2
Copper	4
Drinking Water Best Management Practices	
Bacteria	5
Lead	7
Additional Considerations	12
Resources	14
State Drinking Water and Lead Prevention Information Sources	14
Glossary	15

What Decision Makers Should Know

On any given day in America nearly 50 million public school students spend a significant portion of their day in school buildings. Exposure to environmental hazards in schools can negatively impact the health of children and school staff. Moreover, studies have shown that poor indoor environments in schools have negative impacts on teacher productivity and student performance.^{1,2,3}

Schools and child care facilities receive their drinking water from nearby municipal water systems or their own on-site water system. Facilities that receive their water from a water provider can be assured that the water is regularly tested to ensure it meets federal and state drinking water standards, such as for bacteria and chemicals.

Even if the water meets federal and state standards at the municipal water supply, the water pipes and plumbing fixtures in your facility can affect the quality of water. Therefore, it is important to perform routine measures to limit bacteria or lead contamination and to implement other best management practices for drinking water to ensure your school or child care facility is providing safe drinking water to students and staff.

This guide identifies best management practices for drinking water in schools and child care facilities that are served by municipal drinking water systems to address the following issues:

- Bacteria that may grow within the plumbing system and hot water tanks, on water fountains and faucets, or enter the facility's distribution system (the building's pipes and plumbing) through cross connections;
- Elevated lead levels as a result of contamination from the facility's plumbing and drinking water fixtures; and
- Additional measures such as water conservation and educating students about drinking water.

Top Five Actions to Protect Drinking Water at Schools and Child Care Facilities

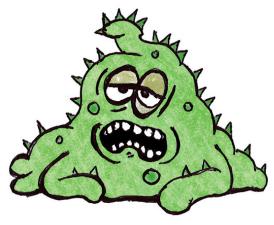
- 1. Clean drinking water fountains daily using procedures found in this guide.
- 2. Clean debris out of all outlet screens or aerators on a regular basis using the procedures found in this guide.
- 3. Test the facility's drinking water for lead. If lead is present, follow the actions for addressing lead contamination outlined in this guide.
- 4. Evaluate your facility for the presence of cross-connections and address any issues related to cross-connections by following the recommendations found in this guide.
- 5. If elevated lead levels are found, regularly flush all water outlets used for drinking or food preparation and install point-of-use devices, to provide additional treatment of drinking water at the outlet.

¹ Vinciullo F. The relationship between multicomponent school health programs and school achievement. Paper presented at: Annual Conference of the National Association of School Nurses, 2008; Albuquerque, NM.

² Stolz A, Knickelbein, A., Coburn, S. Linking coordinated school health to student success. Paper presented at: Annual Conference of the National Association of School Nurses, 2008; Albuquerque, NM

³ NRC (National Research Council). Green Schools: Attributes for Health and Learning. Washington D.C.: The National Academies; 2006.

Bacteria



Bacteria are present throughout our environment. They have adapted to live and reproduce in a variety of environments, including inside animals and humans, and in water, soil and food. If harmful bacteria are present in drinking water sources, most are removed during the disinfection process. However, some may survive and enter the distribution system (the building's pipes and plumbing). Bacteria can also grow within the plumbing system, water fountains and faucets. Therefore, it is important to regularly clean your facility's water fountains, faucets and hot water tanks.

Lead

Understanding Lead Exposure

School officials and child care providers need to know whether the students, teachers and staff consume elevated levels of lead when drinking water in their facility, because exposure to lead can cause serious health problems, particularly for young children.

Lead in drinking water is primarily from materials and components associated with service lines and facility plumbing. Your municipal water supplier is responsible for providing high quality drinking water, but cannot control the variety of materials contained in the plumbing components used within your facility.

Health Effects of Lead

Lead can cause serious health problems if too much enters your body from drinking water or other sources. Some facts about lead exposure include:

- Infants, young children and pregnant women are at greatest risk to lead exposure;
- Increased lead levels have been shown to cause damage to the brain and kidneys;
- Increased lead levels interfere with the production of red blood cells that carry oxygen to all parts of your body;

Reduction of Lead in Drinking Water Act

A new requirement, signed into law by President Obama in January 2011, will further reduce lead in pipes, pipe fittings, plumbing fittings and fixtures to a weighted average of 0.25 percent. The Reduction of Lead in Drinking Water Act redefines "lead free" under the Safe Drinking Water Act to mean: not containing more than 0.2 percent lead when used with respect to solder and flux, the material used to join pipes and fixtures together (current law) and not more than a weighted average of 0.25 percent lead when used with respect to the wetted surfaces of pipes, pipe fittings, plumbing fittings and fixtures. The new requirements will become effective in January 2014.

- Scientists have linked the effects of lead on the brain with lowered intelligence quotient (IQ) in children;
- Adults with kidney problems and high blood pressure can be affected by lower levels of lead more than healthy adults;
- Lead is stored in the bones and it can be released later in life; and
- During pregnancy, the fetus receives lead from the mother's bones which may affect brain development.

How Lead Gets into Drinking Water

Soft water has a low pH, which is corrosive. Other factors however also contribute to the corrosion potential of the water and include water velocity, temperature, alkalinity, the type of disinfectant, the age and condition of plumbing and the amount of time water is in contact with plumbing. Of note, recent construction work on your facility's plumbing system (e.g., pipe replacement and utility lead service line replacement with copper components) may result in corrosion of remaining lead pipes or disturbance of settled debris within larger pipes in the system which may create new sources of contamination. The occurrence and rate of corrosion depend on the complex interaction between a number of these and other chemical, physical and biological factors.

Municipal water systems take steps to reduce the corrosiveness of the water. However, if the plumbing in your facility is made of lead or contains lead parts, corrosion may occur as the water moves through your facility's plumbing.



Example of lead pipes in a plumbing system.

Potential Sources of Lead in Drinking Water

- Lead pipes in plumbing:
 - Dull gray in color and will appear shiny when scratched
 - Banned since 1986 and not widely used since the 1930s
- Copper pipes joined by lead solder:
 - Solder will be dull gray in color and will appear shiny when scratched
 - Banned since 1986 and many communities banned prior to 1986
- Brass pipes, faucets, fittings and valves:
 - May contain alloys of lead
- Sediments in screens on faucets:
 - Debris from plumbing can collect on screens and may contain lead
- Water service line to facility is made of lead:
 - Pipes that carry water from the municipal water system main to the facility may contain lead
- Water fountains in the facility may contain lead parts:
 - Specific brands of water fountains contain lead parts or have lead lined water tanks
 - Since 1988 it has been mandated that water fountains be lead free but older facilities may have outdated models



Copper pipes joined by lead solder.

Copper

Copper is widely used in household plumbing, sometimes without proper consideration of water quality. Excess copper exposure can cause stomach and intestinal distress, liver or kidney damage and complications of Wilson's disease. Children's bodies absorb more lead

and copper than the average adult because of their rapid development. Copper leaches into water through corrosion of the plumbing system – primarily from pipes, but fixtures, faucets and fittings made of brass can also be a source. The amount of copper in your water strongly depends on the acidity and types and amounts of minerals in the water, whether or not it is oxygenated or disinfected, how long the water stays in the pipes, the length of time the pipes have been in use and the water's temperature. Especially when the water pH is below neutral (7) and when the alkalinity of the water (bicarbonate content) is high, very high concentrations of copper can persist for many years in copper pipes and fittings found in new construction and remodeled or renovated buildings. Blue staining of water, sinks and fixtures can be an indicator of extreme copper plumbing corrosion.



Drinking Water Best Management Practices

Bacteria

Drinking Water Fountains

Drinking water fountains should be cleaned on a daily basis to reduce possible bacterial contamination. Fountains should also be included in the regular flushing of the facility's plumbing system (as described in the section Routine Measures for Reducing Lead Exposure).

Drinking Water Fountain Daily Cleaning Procedures¹

The following procedures should be considered for daily cleaning:

- Gather necessary materials and suggested protective equipment;
- Obtain Material Safety Data Sheets (MSDS) for all chemicals being used and review manufacturer's instructions for use;
- Check the flow of the water to make sure there is a constant stream;
- Spray disinfectant cleaner solution on the inside surfaces of the mouthpiece and protective guard;
- Using a scrub brush, clean the inside and outside of the mouthpiece and protective guard;
- Rinse the mouthpiece and protective guard with water; and
- Wipe drinking fountain surfaces with a clean cloth dampened with water.

It is also important to clean drinking water fountains to remove lime and calcium build-up. Lime and calcium build-up can begin to block



Lime build-up on mouthpiece and protective guard of drinking fountain.

the water from coming through the mouthpiece and going down the drain.

Removing Lime Build-up on Drinking Fountains or Ice Machines

- Spray descaler onto the bowl and back of the drinking fountain;
- Use a clean, lint-free cloth saturated with the descaler. Apply to the surfaces with the lime build-up. Let stand for the length of time recommended on the label;
- Wring out all excess solution from the cloth;
- Wipe the surface clean with the cloth. If necessary, use a brush or scrub pad to remove hard build-up. Be careful not to damage surfaces while scrubbing; and
- Thoroughly rinse the surfaces with clean water.

For a list of EPA-approved disinfectants to use in your facility, please visit: http://www.epa.gov/oppad001/chemregindex.htm



¹ Iowa State University Facilities Planning & Management – Custodial Task Procedures

Hot Water Tanks

Hot water tanks are susceptible to the development of biofilm, which is a surface deposit of bacteria that accumulates creating a slime layer. Similar to the plaque that forms on teeth biofilms accumulate over time. It is recommended that you consult with an experienced professional to have your hot water tank periodically cleaned to remove existing biofilms and sediments.²



A cross connection between a dishwasher drain (copper pipe) and a main pipe.

Cross Connections

A cross connection is an actual or potential connection between your facility's drinking water system and other liquids or substances. It is important to be aware of cross connections within your facility, as contamination can occur. Cross contamination from backflow of harmful substances may occur as a result of reduced pressure in the drinking water system or because of increased pressure in the contaminating source. Be aware, cross contamination may not be immediately apparent because a contaminant may not have a strong taste, odor or color.

Boil water orders are issued by your municipal water system when elevated levels of bacteria and viruses are a concern. However, boiling water does not reduce lead levels in drinking water. Because some of the water evaporates during the boiling process, the lead concentration of the water actually increases as the water is heated. Hot tap water also dissolves lead more readily than cold tap water. If you are concerned about lead levels in your water and your municipal water system issues a boil water alert for reducing bacteria or viruses, you should find alternate sources of drinking water until the alert is over.

Cross contamination can happen at facilities under a number of circumstances, including:

- When a tube or hose from a faucet is submerged in a solution, beaker or custodian's sink;
- A pipe is connected from a drinking water source to chemical lab equipment, a storage tank or cafeteria equipment; or
- A hose is dropped into a waste/floor drain in an automotive shop, boiler room or cafeteria.



A garden hose creates a dangerous cross connection between potable and non-potable water.

² National Environmental Services Center, Tech Brief. Biofilm Control in Distribution Systems, Summer 2008, Vol. 8, Issue 2.

Other sources of potential cross contamination include cross connections between the drinking water system and heating system boilers, water coolers, lawn sprinkler systems, fire sprinkler systems or soft drink machines.

Cross contamination can be prevented by using backflow prevention devices that only allow water to flow in one direction, from the source to the tap, so liquid cannot flow back down the tap and contaminate the water in the distribution system. Backflow prevention devices should be tested annually. If you have questions about cross connections and contamination, contact the local building/plumbing inspector or for more information on cross contamination please visit: http://water.epa.gov/infrastructure/drinkingwater/pws/crossconnectioncontrol/index.cfm.

Lastly, inspect your facility's pipes to make sure there are no leaks or outside indications of corrosion. Leaks in your plumbing system can be a way for bacteria to get into your water.

Lead

Voluntary Testing of Lead in Drinking Water

You cannot see, taste or smell lead in your drinking water. As such, testing the facility's water is the only sure way to know if there are elevated levels of lead in the water.

EPA developed the 3Ts for Reducing Lead in Drinking Water in Schools, Revised Technical Guidance (Training, Testing and Telling) to help schools and child care facilities implement simple strategies for managing the health risks of lead in drinking water. The 3Ts include:



Lead exposure poses a great risk to young children.

- Training school and child care facility officials to raise awareness of the potential occurrences, causes and health effects of lead in drinking water, assist facilities in identifying potential areas where elevated lead may occur, and establish a testing plan to identify and prioritize testing sites;
- Testing drinking water in the facility to identify potential problems and take corrective actions as necessary; and
- Telling students, parents, staff and the larger community about monitoring programs, potential risks, the results of testing and remediation actions.

The 3Ts for Reducing Lead in Drinking Water in Child Care Facilities is available at: http://water.epa.gov/infrastructure/drinkingwater/schools/guidance.cfm#3ts

EPA also developed the, "What Your School or Child Care Facility Should Know About Lead in Drinking Water" DVD available for order from the National Service Center for Environmental Publications (NSCEP) at: http://water.epa.gov/infrastructure/drinkingwater/schools/guidance.cfm³

Test the Facility's Drinking Water for Lead



It is important to test all of the drinking water outlets in your facility, especially those that provide water for drinking, cooking and preparing juice and infant formula. Lead in drinking water can be a very localized problem and can vary from tap to tap. Just because there is lead getting into your water from one outlet does not mean that all your taps are vulnerable or contaminated. At the same time, just because one tap sample is free from lead does not mean that all your taps are clear. It is a good idea to test ALL outlets including drinking fountains and water faucets where water will be used for drinking or cooking. Unusual sources of drinking water, such as locker room shower heads and other nondrinking water taps used to fill water jugs and carboys, should also be included when tested for lead.

If your sink has separate hot and cold water knobs then samples should be collected from cold water as hot tap water is not recommended for food preparation or direct consumption. If you have one lever, be sure to turn it on to the cold water side. The 3Ts for Reducing Lead in Drinking Water in Schools, Revised Technical Guidance, provides step-by-step guidance on how to properly collect samples and test your facility's drinking water outlets for lead.⁴

The concentration of lead in your drinking water samples will be reported in metric form, such as milligrams per liter (mg/L) or micrograms per liter (µg/L), or as parts per million (ppm) or parts per billion (ppb).

EPA recommends that schools and child care facilities take action if samples from any drinking water outlets show lead levels greater than 20 parts per billion.

Routine Measures for Reducing Lead Exposure

Whether you have tested your water or not, or even if you have and your water has shown low levels of lead, there are best practices that will further reduce the potential for lead exposure at your facility as well as reduce sediment in your water.

Develop a flushing plan

- Determine how water enters and flows through your facility by developing a plumbing profile.⁵ Consult with your maintenance personnel, licensed plumber or local water service provider to develop a plumbing profile;
- Locate all water outlets that are used for consumption;

³ Also available by calling NSCEP at 1-800-490-9198. For International Orders: Call NSCEP at (301) 519-6640 or e-mail NSCEP at nscep@bpslmit.com

⁴ See Section II: Testing of the 3Ts for Reducing Lead in Drinking Water in Schools, Revised Technical Guidance, available at: http://water.epa.gov/ infrastructure/drinkingwater/schools/guidance. cfm#3ts

⁵ See Section II: Testing of the 3Ts for Reducing Lead in Drinking Water in Schools, Revised Technical Guidance, available at: http://water.epa.gov/ infrastructure/drinkingwater/schools/guidance. cfm#3ts

- Identify the drinking water outlet(s) farthest from the main water service line (Note: If your facility has multiple wings there will be more than one outlet farthest from the main service line);
- Determine the best order to open and flush drinking water outlets, starting with those farthest from the main service line;
- · Identify options for collection and nonpotable re-use of flushed water (e.g., plant watering); and
- Develop a system for accountability, including identifying one person who is in charge and developing a record keeping system.

Flush all water outlets used for drinking or food preparation

- At the start of each day, before using any water for drinking or cooking, flush the cold water faucet by allowing the water to run for a period of time. Contact your municipal water system to find out what the recommended flushing time is for your facility based on your system size and piping. This should be done for all water outlets used for drinking or food preparation.
- Flushing, or opening up a tap and letting the water run, replaces the stagnant water that may have been in contact with leadcontaining plumbing fixtures overnight or over the weekend. The longer water is exposed to lead pipes or solder the greater the likelihood of lead contamination.
- Flushing times vary depending on your buildings pipes and outlets, refrigerated water fountains can take as long as 15 minutes to properly flush out the reservoir.
- If many taps need flushing, the tap farthest from the main pipe should be opened for approximately ten minutes to flush out the main pipe. Then, individual drinking water taps should be flushed to rid stagnant water from the pipes. Keep in mind that if your facility has more than one wing there may be

more than one tap that is furthest from the water line.6

Use only cold water to prepare food and drinks

- Hot water dissolves lead more quickly than cold water and is therefore more likely to have greater amounts of lead.
- If hot water is needed, water should be drawn from the cold tap and heated.
- Use only thoroughly flushed water from the cold water tap for drinking and when making mixed baby formula, juices or foods.

Clean debris out of all outlet screens or aerators on a regular basis

• Small screens on the end of a faucet can trap sediments containing lead. Note: Aerators are often used to regulate flow, reduce splash and conserve water. Check to see if your faucets have aerators, since not all faucets have them.



Faucet Aerators

Many taps that are used to provide water for human consumption have an aerator as part of the faucet assembly. Aerators serve to introduce air into the water flow which makes it feel as if a larger water flow is coming out of the tap. The use of aerators is a common water conservation practice. Screens are not intended to remove contaminants in the water, but may trap sediment or debris as water passes through the faucet. Lead bearing sediment may end up in drinking water from physical corrosion of leaded solder and can build up in the aerator over time.

⁶ Lead in School Drinking Water Program (http:// www.mass.gov/dep/water/drinking/sclcatlg.pdf).

Faucet Aerators Cleaning Procedures

• Remove the aerator by twisting off with hands or pliers;



 One or more parts are contained within the aerator. Note the order and orientation of the parts as you remove them;



 Rinse the pieces with water and brush off the debris. For deposits that are difficult to remove, soak the parts in water for a few minutes and scrub with a new toothbrush.
 Backwashing aerator components is also an effective cleaning method for many aerator types. Hold removed aerator upside down under flowing water to backwash screens and mesh filters;



• If any parts are cracked or broken, replace them. If the washer has hardened it should be replaced; and



• Reassemble the aerator by screwing it back onto the faucet and hand-tighten.

It is not recommended that aerators be removed from faucets immediately before sampling for lead as the sample will fail to identify the typical available contribution of particulate lead from that tap and thus additional actions needed to reduce exposure to lead in drinking water will fail to be taken.

However, if the results from the initial sample are above the action level, you can consider taking a second sample to determine whether particulate matter is the source of lead. For this sample, the aerator would be cleaned or removed prior to sampling so that the two samples could be compared.⁷ It is advised that a regular cleaning schedule be established for aerators.

The frequency of cleaning depends on how quickly the debris builds up on the inside walls of the aerator. Generally, the aerator should be cleaned at least quarterly. If blockages from scale or particulate accumulation are noticed, the frequency should be increased to monthly. Flow pattern changes are an indication of buildup.

⁷ EPA, Oct 2006. Memorandum: Management of Aerators during Collection of Tap Samples to Comply with the Lead and Copper Rule. Available at: http://water.epa.gov/lawsregs/rulesregs/sdwa/lcr/upload/2006_10_27_lcrmr_memo_tapsamplesaerators_10202006.pdf.

Follow-up lead monitoring

Water quality changes over time and it is important to make sure that the water in your facility remains free of lead. Periodic retesting of fixtures is recommended.

Respond to Elevated Lead Levels

EPA recommends that schools and child care facilities take action if samples from drinking water outlets show lead levels greater than 20 ppb. Any outlet with test results above this level should not be used until the source of the contamination is found and the lead levels are reduced to 20 ppb or less. If you are going to stop using an outlet due to high lead levels you should place a physical barrier, such as tape or an illustrative sign over the faucet so that everyone knows it should not be used until it is fixed. In addition, you should encourage parents to have their children's blood tested for lead if high lead levels are detected in the water. Facilities should develop Standard Operating Procedures (SOPs) for responding to elevated lead levels and administrators or directors should be encouraged to communicate this information to parents (or the teachers) so they can protect their children. The following remedies can be used to respond to elevated lead levels:

Provide an alternative "lead-free" drinking water

- Bottled water can be used as a temporary measure; and
- Make sure the bottled water distributor meets federal and state bottled water quality standards (which are different than tap water) and that their filtration technology is National Sanitation Foundation International (NSF) certified for lead reduction (http:// www.nsf.org/).

Remove sources of lead in the plumbing system

These remedies are most appropriate for localized contamination problems and are best handled by a licensed plumber:

- Replace solder joints with lead-free joints;
- Replace the outlet or fixture/faucet with "lead-free" materials (according to NSF/American National Standards Institute (ANSI) Standards 61 and 372; or
- Replace piping with "lead-free" materials (according to NSF/ANSI Standards 61 and 372).

You can search for NSF Drinking Water System Components here: http://www.nsf.org/ certified/pwscomponents/

Prior to replacing fixtures when elevated lead levels are determined, be sure to test the new fixtures to ensure the fixtures are "lead free." If you are purchasing a large volume of faucets ask the manufacturer or vender to test the faucets with your local tap water to make sure no lead is leached out of the faucets. If you are only purchasing a few faucets make sure the fixtures are certified as lead-free according to NSF/ANSI Standard 61 for lead content.

Install point-of-use treatment devices



A faucett filtration system.

- A point-of-use (POU) device is a filtration system that can be installed directly on a drinking water outlet.
- Use a device that is certified by NSF

- International to remove lead.8
- Maintaining POU treatment devices is very important. Refer to the manufacturer's instructions for maintenance procedures. If not maintained properly, some treatment devices may increase lead and other contaminant levels.
- If using a POU device you should do follow up testing to make sure the water is still below the action level.
- With the use of a POU device, flushing is not necessary.
- If using POU devices on some faucets, but not all, make sure that faucets without a POU device are clearly labeled that they are not for drinking or cooking water.

Communicate with Your Community about the Voluntary Lead Testing Program

It is important to communicate early and often about your testing plans, results and next steps. Telling parents and staff about your voluntary lead testing program will demonstrate your proactive commitment to protecting the health of your students and staff and build confidence in your facility's ability to provide a safe and healthy environment, whether or not elevated lead levels are found in your facility.

Additional Considerations

Copper

Testing for copper may be appropriate if your water is somewhat acidic (with a pH below 7) and when it is disinfected. Copper corrosion decreases steadily over time under normal water usage conditions, but elevated copper levels can persist for many years in new copper pipes. If you are experiencing blue staining of your water, sinks, bathtubs or showers or if there is growing blue encrustation on the fixtures, this may be an indicator of high copper levels, and you should have your water tested for both copper and lead. If you are experiencing elevated copper levels in the drinking water system, the easiest method for reducing exposure to copper is to flush the system to avoid drinking or cooking with water that has been in contact with the plumbing system for more than four hours. Particularly when first drawing water in the morning, flush the system by running the cold water faucet long enough to get fresh water from the main, which could vary from about two minutes to five or ten minutes, depending on the size and length of the pipes and the flow rate. Each faucet where water is drawn for drinking or cooking purposes should be flushed separately, starting with the faucet or outlet farthest from where the fresh water enters the building.

⁸ The Lead and Copper Rule: Quick Reference Guide for Schools and Child Care Facilities that are Regulated Under the Safe Drinking Water Act, available at: http://water.epa.gov/infrastructure/drinkingwater/schools/upload/2006_1_11_schools_lead_sqrg_lcr_schools.pdf.

Water Conservation

Schools and child care facilities use large amounts of water every day for heating and cooling systems, restrooms, drinking water, cooking, locker rooms, cafeterias, laboratories and outdoor playing fields and lawns. Options to conserve water for these facilities include:

- Consider replacing old equipment such as dishwashers with energy and water saving devices;
- Repair water leaks and leaky toilets;
- Install aerators and automatic shut-off devices on faucets;
- Using low-flow shower heads and timer shut-off devices to reduce water use during showers;
- Installing timers on sprinklers;
- Installing toilet dams on older models;
- Replacing plants and grasses that require a lot of water with native sustainable vegetation; and
- Using rain barrels and teach the students about green roofs.

WaterSense

The WaterSense program promotes water efficiency and enhancing the market for water-efficient products, programs and practices. Since the program's inception in 2006, WaterSense has helped consumers save a cumulative 46 billion gallons of water and \$343 million in water and sewer bills. For additional information, visit: http://www.epa.gov/watersense/

Teaching Students about Drinking Water

There are a variety of ways to teach students about drinking water:

- Early science classes demonstrating the water cycle;
- Mathematics classes demonstrating supply and demand principles;

- History lessons discussing early settlement patterns near water sources and our nation's system of government, laws and regulations provide important knowledge for water resource decision-making.
- Involve students in the voluntary lead testing program to make it a teaching moment, ensuring that the students wear proper safety equipment and are not exposed to lead.

EPA has developed numerous resources and activities for students and teachers, including:

Drinking Water in Schools and Child Care Facilities (EPA)

Provides multiple resources regarding lead in drinking water, including health information, an overview of laws and regulations and guidance. Website: http://water.epa.gov/infrastructure/drinkingwater/schools/index.cfm

Kid's Stuff: Drinking Water and Ground Water (EPA)

Provides activities and materials for students and teachers for grades K-3, 4-8 and 9-12. Website: http://water.epa.gov/learn/kids/drinkingwater/index.cfm

Water Science and Technology for Students and Educators (EPA)

Water-related activities and resources for students and teachers.

Website: http://water.epa.gov/learn/resources/index.cfm

That Magnificent Ground Water Connection (New England Interstate Water Pollution Control Commission in coordination with EPA)

Classroom activities for students demonstrating the many characteristics, uses and threats to ground water resources in New England.

Available for grades K-6 and 7-12.

Selected activities are available at: http://www.epa.gov/region1/students/teacher/groundw.html

Resources

3Ts for Reducing Lead in Drinking Water in Schools

Provides detailed guidance for schools that receive their drinking water from municipal water supplies regarding training and testing for and communicating about lead in drinking water.

Website: http://water.epa.gov/infrastructure/drinkingwater/schools/guidance.cfm#3ts

3Ts for Reducing Lead in Drinking Water in Child Care Facilities

Provides detailed guidance for child care facilities that receive their drinking water for municipal water supplies regarding training and testing for and communicating about lead in drinking water.

Website: http://water.epa.gov/infrastructure/drinkingwater/schools/guidance.cfm#3ts

EPA's Website on Lead

http://www.epa.gov/lead/

EPA's Website on Lead in Drinking Water http://water.epa.gov/drink/info/lead/index.

cfm

EPA's Website on Reducing Lead in Drinking Water in Schools and Day Care Centers

http://water.epa.gov/drink/info/lead/schools_index.cfm

Centers for Disease Control and Prevention's Website on Lead

http://cdc.gov/lead/

National Lead Information Center Hotline: (800) 424-LEAD

EPA's Safe Drinking Water Hotline:

(800) 426-4791

Are You Providing Safe Drinking Water at Your School or Child Care Facility?

For Schools and Child Care Facilities With Their Own Drinking Water Source

Caring for Our Children: National Health and Safety Performance Standards Guidelines for Early Care and Education Programs, 3rd Edition (2011)

Provides national standards that represent the best evidence, expertise and experience in the country on quality health and safety practices and policies that should be followed in today's early care and education settings. Chapter 5 covers drinking water safety, lead testing and appropriate plumbing.

Website: http://nrckids.org/CFOC3/index. html

State Drinking Water and Lead Prevention Information Sources

State drinking water programs can describe state-specific requirements and provide additional guidance materials for schools. For a complete list of State Drinking Water program contacts and lead prevention information sources, see:

Implementing the Lead Public Education Provision of the Lead and Copper Rule for Non Transient, Non Community Water Systems, Appendix C:

http://water.epa.gov/lawsregs/rulesregs/sdwa/lcr/upload/Implementing-the-Lead-Public-Education-Provisions-of-the-Lead-and-Copper-Rule-A-Guide-for-Non-Transient-Non-Community-Water-Systems.pdf

Implementing the Lead Public Education Provision of the Lead and Copper Rule for Community Water Systems, Appendix C:

http://water.epa.gov/lawsregs/rulesregs/sdwa/lcr/upload/Implementing-the-Lead-Public-Education-Provisions-of-the-Lead-and-Copper-Rule-A-Guide-for-Community-Water-Systems.pdf

Glossary

Acidic:

The condition of water or soil which contains a sufficient amount of acidic substances to lower the pH below 7.0.

Action Level:

The level of lead or copper which, if exceeded, triggers treatment or other requirements that a water system must follow.

Alkalinity:

The capacity of water to neutralize acids. This capacity is caused by the water's content of carbonate, bicarbonate, hydroxide and occasionally borate, silicate and phosphate. Alkalinity is expressed in milligrams per liter of equivalent calcium carbonate. Alkalinity is not the same as pH because water does not have to be strongly basic (high pH) to have a high alkalinity. Alkalinity is a measure of how much acid can be added to a liquid without causing a significant change in pH.

Alloy:

A solution made of two or more elements, at least one of which is a metal.

Backflow:

A reverse flow condition created by a difference in water pressures which causes water to flow back into the distribution pipes of a potable water supply from any source or sources other than an intended source.

Backwashing:

The process of reversing the flow of water back through the filter media to remove the entrapped solids.

Bacteria:

Microscopic living organisms usually consisting of a single cell. Bacteria can aid in pollution control by consuming or breaking down organic matter in sewage or by similarly acting on oil spills or other water pollutants. Some bacteria in soil, water or air may also cause human, animal and plant health problems.

Contaminant:

Anything found in water (e.g., microorganisms, minerals, chemicals, radionuclides, etc.) which may be harmful to human health.

Corrosion:

The gradual decomposition or destruction of a material by chemical action often due to an electrochemical reaction. Corrosion may be caused by: 1) stray current electrolysis, 2) galvanic corrosion caused by dissimilar metals or 3) differential concentration cells. Corrosion starts at the surface of a material and moves inward.

Cross-Connection:

Any actual or potential connection between a drinking (potable) water system and an unapproved water supply or other source of contamination. For example, if you have a pump moving non-potable water and hook into the ground water system to supply water for the pump seal a cross-connection or mixing between the two water systems can occur. This mixing may lead to contamination of the drinking water.

Descaler:

A solution used to remove and/or prevent limescale and fouling on water taps, kettles, coffeemakers, toilets and water pipes.

Disinfectant:

A chemical (commonly chlorine, chloramine or ozone) or physical process (e.g., ultraviolet light) that kills microorganisms such as bacteria, viruses and protozoa.

Distribution System:

A network of pipes leading from a treatment plant to customers' plumbing systems or the pipes and plumbing within a building that distribute water to all of the water outlets.

Ground Water:

The water that systems pump and treat from aquifers (natural reservoirs below the earth's surface).

Lead Service Line:

A service line made of lead which connects the water main to the building inlet and any lead pigtail, gooseneck or other fitting which is connected to such a lead line.

Monitoring Program:

Testing that water systems must perform to detect and measure contaminants. Specifically, measuring concentrations of certain substances within environmental media (e.g., drinking water) at regularly scheduled intervals.

Municipal Water System:

A network of pipes, pumps and storage and treatment facilities designed to deliver potable water to homes, schools, businesses and other users in a city or town.

Non-Potable Water:

Water that may contain objectionable pollution, contamination, minerals or infective agents and is considered unsafe and/or unpalatable for drinking.

pH:

A measurement of how acidic or basic a substance is. It ranges from 0 to 14. A pH of 7 is neutral. A pH less than 7 is acidic and a pH greater than 7 is basic.

Point-of-Use Device:

A treatment device applied to a single tap used for the purpose of reducing contaminants in drinking water at that one tap.

Potable Water:

Water that is safe and satisfactory for drinking and cooking.

Remediation:

Removal of pollution or contaminants from environmental media such as soil, ground water, sediment or surface water for the general protection of human health and the environment.

Samples:

The water that is analyzed for the presence of EPA-regulated drinking water contaminants. Depending on the regulation, EPA requires water systems and states to take samples from source water, from water leaving the treatment facility or from the taps of selected consumers.

Soft Water:

Water having a low concentration of polyvalent cations, such as calcium and magnesium ions. According to U.S. Geological Survey guidelines, soft water is water having a hardness (concentration of polyvalent cations) of 60 milligrams per liter or less.

Solder:

A metallic compound used to seal the joints between pipes. Until recently, most solder contained 50% lead. The use of lead solder containing more than 0.2% lead is now prohibited for pipes carrying potable water.

Toilet Dam:

A water-conservation device that is placed inside the tank portion of a toilet to reduce the amount of water the tank will hold by partitioning off part of the tank.